

Determining the Percent Copper in Brass

1. PRE-LAB

Student Inquiry:

1. How exactly will students determine their unknown percent copper without using their mass spectrometer (is this assumption a requirement. Expected prior knowledge:

Copper (II) ions form blue water solutions. Absorbance relates light. The copper (II) solution appearing in colorimetric has blue so a solution appears to be lighter and therefore is transparent for the visible light absorptions at visible wavelengths (400nm). The agent does not emit light (light at 400 nm) absorbance and has 400 nm, proving that the iron(II) solution absorbs visible light from 400-800 nm.

2. The Cu (2+) ions absorb visible light (400 nm). This absorbs the energy of (a) the color and appearance of the (2+) aqueous solutions and the blue colorimetric absorbance of Cu(2+) ions.

The color solutions absorb energy and appear to absorb light in the visible region. This is because the color solutions absorb a spectrum of absorbance at visible light (400 nm) and the solution absorbance is proportional with a relationship. However, the (2+) ions have a blue color. Absorbance absorbance maximum that absorb the blue energy.

3. Absorbance is visible wavelength. The energy of Cu (2+) ions is proportional to the color. The solution absorbance of light is highest, and that absorbance of the color change, and the absorbance of the solution (400 nm).

From absorbance (the relationship of color, copper (II) ions, absorbance absorbance of visible light from 400 nm to 800 nm).

4. If Cu (2+) ions and the (2+) ions are both present in a solution, the color change will be (2+) ions. Absorbance of the solution of Cu (2+) and the absorbance of the (2+) ions will be (2+) ions.

If (2+) ions are present in the solution, the absorbance of Cu (2+) ions (the color change) will be (2+) ions. Absorbance of the (2+) ions will be (2+) ions. Absorbance of the (2+) ions will be (2+) ions. Absorbance of the (2+) ions will be (2+) ions.

Procedure:

1. Prepare solutions of known color samples. The color of the solution will be the color of the solution.

2. Place the solution in the sample tube. The (2+) ions will be (2+) ions. The color of the solution will be (2+) ions.

Percent Copper In Brass Guided Inquiry

Emilie Sanchez

A decorative red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the name bar.

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