Name:

 Circle the element with the largest atomic radius and put a square around the element with the smallest atomic radius:

Cu K Ni Br

- a. Explain why you made these choices: All of the elements are in the same period. The trend in atomic radius as you go across a period is DECREASING. Therefore, the element on the far left (K) is the largest, and the element on the far right (Br) is the smallest.
- Circle the element with the highest ionization energy and put a square around the element with the lowest ionization energy:

Cu K Ni (Br)

- a. Explain why you made these choices: All of the elements are in the same period. The trend in ionization energy as you go across a period is INCREASING. Therefore, the element on the far left (K) has the lowest ionization energy, and the element on the far right (Br) has the highest ionization energy.
- Circle the element with the highest electronegativity and put a square around the element with the lowest electronegativity:

Cu K Ni Br

- a. Explain why you made these choices: All of the elements are in the same period. The trend in electronegativity as you go across a period is INCREASING. Therefore, the element on the far left (K) has the lowest electronegativity, and the element on the far right (Br) has the highest electronegativity.
- 4. For each of the following groups: Circle the element with the largest atomic radius and put a square around the element with the smallest atomic radius:

5.

- a. O C (Bc) Nc Same Period
 b. Na Rb (Fr) H Same Group
 c. (Pb) C Sn Si Same Group
 d. Au W S (Fr) Nc Zn Challenge
- For each of the following groups: Circle the element with the highest ionization energy and put a square around the element with the lowest ionization energy:
 - a. O C Bc No Same Period
 b. Na Rb Fr H Same Group
 c. Pb C Sn Si Same Group
 d. Au W S Fr No Zn Challenge

Periodic Table Trends Test Answer Key

Marco Cascella

Periodic Table Trends Test Answer Key:

Decoding **Periodic Table Trends Test Answer Key**: Revealing the Captivating Potential of Verbal Expression

In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Periodic Table Trends Test Answer Key**," a mesmerizing literary creation penned with a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring impact on our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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