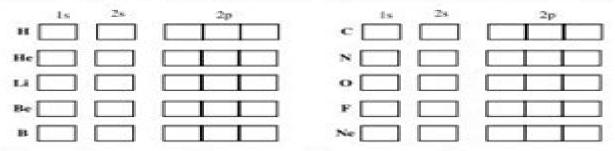
POGIL: Periodic Trends in Atomic Properties

Information

Properties such as the size of an atom (atomic radius), the energy required to remove an electron from an atom (ionization energy), and the energy associated with the addition of an electron to a gaseous atom (electron affinity) can be understood in terms of the electron configuration of the atom and a competition between electron-nucleus attraction and electron-electron repulsion.

Preliminary Activity - Complete the orbital diagrams for the ground state configuration.



Model 1: Electron Ionization & Capture and Electrostatic Interactions

Reaction Equation for Electron Ionization:

 $X \rightarrow X^* + e^-$

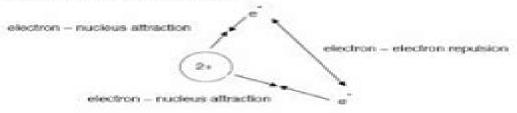
 $\Delta E = 1^{a}$ ionization energy

Reaction Equation for Electron Capture:

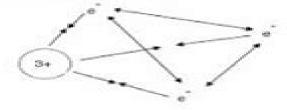
 $X + e \rightarrow X$

 $\Delta E = electron affinity$

Dominant inter-particle forces in helium



Dominant inter-particle forces in lithium:



Periodic Trends And Atomic Properties Pogil Answers

Alex Kimmons

Periodic Trends And Atomic Properties Pogil Answers:

Atoms, Molecules & Elements Gr. 5-8 George Graybill, 2007-09-01 Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource makes the periodic table easier to understand Begin by answering what are atoms See how the atomic model is made up of electrons protons and neutrons Find out what a molecule is and how they differ from elements Then move on to compounds Find the elements that make up different compounds Get comfortable with the periodic table by recognizing each element as part of a group Examine how patterns in the period table dictate how those elements react with others Finally explore the three important kinds of elements metals nonmetals and inert gases Aligned to the Next Generation Science Standards and written to Bloom's Taxonomy and STEAM initiatives additional hands on experiments crossword word search comprehension quiz and answer key are also included **The Periodic Table** Eric R. Scerri, 2019-07-25 The periodic table of elements first encountered by many of us at school provides an arrangement of the chemical elements ordered by their atomic number electron configuration and recurring chemical properties and divided into periodic trends In this Very Short Introduction Eric R Scerri looks at the trends in properties of elements that led to the construction of the table and shows how the deeper meaning of the table's structure gradually became apparent with the development of atomic theory and in particular quantum mechanics which underlies the behaviour of all of the elements and their compounds This new edition publishing in the International Year of the Periodic Table celebrates the completion of the seventh period of the table with the ratification and naming of elements 113 115 117 and 118 as nihonium moscovium tennessine and oganesson Eric R Scerri also incorporates new material on recent advances in our understanding of the origin of the elements as well as developments concerning group three of the periodic table ABOUT THE SERIES The Very Short Introductions series from Oxford University Press contains hundreds of titles in almost every subject area These pocket sized books are the perfect way to get ahead in a new subject quickly Our expert authors combine facts analysis perspective new ideas and enthusiasm to make interesting and challenging topics highly readable **Chemical Periodicity** Robert Thomas Sanderson, 2013-04-20 Chemical Elements Pocket Guide Coventry House Publishing, 2020-02-08 The Chemical Elements Pocket Guide serves as a portable reference for quick study and efficient review of the 118 elements on the periodic table This on the go resource details the physical and atomic properties of each element as well as their history and characteristics in bullet point format The book s small trim size 4 25 x 6 8 inches is intended to fit inside a lab coat pocket and the bound design means you no longer need to carry loose bulky flashcards that can be misplaced or destroyed Includes the updated names nihonium moscovium tennessine and oganesson for elements 113 115 117 and 118 respectively Information provided includes Atomic number Atomic symbol Element category Standard state Atomic mass Electron configuration Oxidation states Electronegativity Atomic radius Ionization energy Electron affinity Melting point Boiling point Density Year discovered Discovered by Appearance Natural occurrence Interesting fact Science Spectrum Holt

Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004 The Amazing World of the Atom and the Ultimate Periodic Table of the Elements Costel Rizescu, Mihaela Rizescu, 2018-05-24 This book is addressed to undergraduate and graduate students taking physics and chemistry courses at an introductory level STEM high school students and to more experienced professors and researchers who need to stay up to date with the most recent advances in data as well It is also beneficial to those involved in materials science medical physics biotechnology astronomy quantum chemistry Earth and space science and other related fields The volume consists of two parts the first part is devoted to the quantum mechanics theory of atomic structures as discovered and verified by physics whereas the second part presents and analyzes a large amount of the latest critically evaluated data for the properties of all elements on the periodic table of the elements Periodic trends for ionization energy electronegativity atomic radius density and melting points are graphed and thoughtfully analyzed Additional data of ionic radius covalent radius van der Waals radius and 12 coordination atomic radius for almost all elements on the periodic table are also presented and evaluated The authors discussed over 30 problems review questions and their solutions to help deepen the insight in this subject area to make this book a real study text Plenty of problems are given to elucidate the material This book represents the culmination of the authors many years of research and teaching physics materials science and instrumentation It is also a crystallization of their passion and strong interest in the history of atomic physics and the philosophy of science in general From the beginning it was authors intention to include information in this book that is not easily located elsewhere or is not found completely in a single book or publication Thus while this book can be used as a text the authors hope that it will be used as a useful reference too Each chapter and sub chapter of this book capitalizes on the strengths comments feedback and criticism that the authors expect to have from students faculty and working professionals

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