

MYP4 Chemistry

Periodic Trends

Use the periodic table, charts, and your knowledge of periodic trends to answer the following questions.

Atomic Radius

1. Which atom in each pair has the larger atomic radius? Circle your answer.

a) Li or **K** b) **Ca** or Ni c) **Ga** or B d) O or **C** e) Cl or **Br**
f) Be or **Ba** g) **Si** or S h) Fe or **Au**

2. What is the periodic trend for atomic size from top to bottom in a group? From left to right in a period?

Atomic size (atomic radius) increases from top to bottom in a group and decreases from left to right in a period.

3. Why do atoms get smaller as you move left to right in a period?

Atoms get smaller as you move left to right in a period due to the increase in effective nuclear charge (while the number of occupied energy levels remain constant) leading to an increase in the force of attraction, which pulls the outer electrons closer to the nucleus and decreases the atomic radius.

4. Why do atoms get larger as you move down a group?

Atoms get larger as you move down a group due to the increase in occupied energy levels.

5. Arrange the following in order of increasing atomic radius atomic radius.

Mg, Na, P, Si and Ar? They are in the same period: Ar, P, Si, Mg, Na

Ionic Radius

6. Define an ion.

a charged atom or group of atoms

7. What is the difference between a cation and an anion?

Cations are positively charged ions. They are formed when an atom loses electrons.
Anions are negatively charged ions. They are formed when an atom gains electrons.

8. Explain the relationship between the relative size of an ion to its neutral atom and the charge on the ions.

Cations are smaller than their neutral atom because they have less occupied energy levels.
Anions are larger than their neutral atom because of the increase in electron-electron repulsion.

9. Which particle has the larger radius in each atom/ion pair? Circle your answer.

a) **Na** Na⁺ b) S, **S²⁻** c) I, I⁻ d) **Al** Al³⁺

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A Gutmann



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Homenatge professor Josep M. Costa (eBook) 1a part. Trends in electrochemistry and corrosion at the beginning of the 21st century Pere-Lluís Cabot Julia, Enric Brillas Coso, 2004-04-13 Esta primera parte del libro Trends in Electrochemistry and Corrosion at the beginning of the 21st century dedicado al Prof Josep M Costa en ocasi n de su 70 aniversario re ne un total de 33 art culos y revisiones originales tanto cient ficas como tecnol gicas correspondientes al campo de la Electroqu mica Estos trabajos est n escritos en espa ol e ingl s por unos 110 investigadores de todo el mundo y son exponentes representativos del gran nivel alcanzado por la investigaci n internacional en diversas materias de gran inter s en la Electroqu mica de principios de este siglo XXI Los trabajos se han agrupado en 5 cap tulos generales relativos a los campos de Electroqu mica Molecular Electrodeposici n Electrodos Modificados Descontaminaci n Electroqu mica y Sensores y Electroan lisis This first part of the book Trends in Electrochemistry and Corrosion at the beginning of the 21st century dedicated to Professor Josep M Costa in occasion of his 70th birthday collects 33 original papers and reviews both scientific and technologic corresponding to the field of Electrochemistry These works are written in English and Spanish by about 110 researchers of all around the world and are representative exponents of the high level reached by the international research in several topics of great interest in Electrochemistry at the beginning of the 21st Century The works have been distributed into five general chapters related to the fields of Molecular Electrochemistry Electrodeposition Modified Electrodes Electrochemical Depollution and Sensors and Electroanalysis Inorganic Chemistry Tina Overton, Jonathan Rourke, Fraser A. Armstrong, 2018 Leading the reader from the fundamental principles of inorganic chemistry right through to cutting edge research at the forefront of the subject Inorganic Chemistry Seventh Edition is the ideal course companion for the duration of a student s degree The authors have drawn upon their extensive teaching and research experience to update this text the seventh edition retains the much praised clarity of style and layout from previous editions while offering an enhanced section on expanding our horizons The latest innovative applications of green chemistry have been added to clearly illustrate the real world significance of the subject This edition also sees a greater used of learning features including substantial updates to the problem solving questions additional self tests and walk through explanations which enable students to check their understanding of key concepts and develop problem solving skills Providing comprehensive coverage of inorganic chemistry while placing it in context this text will enable the reader to fully master this important subject Online Resources Inorganic Chemistry Seventh Edition is accompanied by a range of online resources For registered adopters of the text DT Figures marginal structures and tables of data ready to download DT Test bank For students DT Answers to self tests and exercises from the book DT Tables for group theory DT Web links DT Links to interactive structures and other resources on www.chemtube3d.com **Scientific and Technical Aerospace Reports** ,1984 Lists citations with abstracts for aerospace related reports obtained from world wide sources and announces documents that have recently been entered into the NASA

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