

CHAPTER 6 STUDY GUIDE

The Periodic Table and Periodic Law

Section 6.1 Development of the Modern Periodic Table

In your textbook, reads about the history of the periodic table's development.

Use each of the terms below just once to complete the passage.

octaves	atomic mass	atomic number	size
elements	properties	Henry Moseley	eight
protons	periodic law	Dmitri Mendeleev	<u>accepted</u>

The table below was developed by John Newlands and is based on a relationship called the law of (1) properties. According to this law, the properties of the elements repeated every (2) eight elements. Thus, for example, element two and element (3) nine have similar properties. The law of octaves did not work for all the known elements and was not generally (4) accepted.

1	2	3	4	5	6	7
H	Li	G	Be	C	N	O
8	9	10	11	12	13	15
F	Na	Mg	Al	Si	P	S

The first periodic table is mostly credited to (5) Mendeleev. In his table, the elements were arranged according to increasing (6) atomic mass. One important result of this table was that the existence and properties of undiscovered (7) elements could be predicted.

The element in the modern periodic table are arranged according to increasing (8) atomic number, as a result of the work of (9) Moseley. This arrangement is based on number of (10) protons in the nucleus of an atom of the element. The modern form of the periodic table results in the (11) periodic law which states that when elements are arranged according to increasing atomic number, there is a periodic repetition of their chemical and physical (12) properties.

Periodic Table And Periodic Law Answers

Lakhmir Singh & Manjit Kaur



Periodic Table And Periodic Law Answers:

CBSE Chemistry Chapterwise Case Study Class 11 Priti Singhal, 2024-11-17 This book is structured to align with the latest syllabus and curriculum guidelines ensuring that the content is both relevant and rigorous Each chapter begins with a clear set of learning objectives providing a roadmap for students to understand what they will achieve by the end of the chapter We have included numerous diagrams illustrations and real life examples to make complex concepts more accessible and engaging , *Most Likely Question Bank - Chemistry: ICSE Class 10 for 2022 Examination* Oswal Publishers, 2021-05-15 Benefit from Category wise Chapterwise Question Bank Series for Class 10 ICSE Board Examinations 2022 with our Most Likely ICSE Question Bank for Chemistry Subjectwise book dedicated to prepare and practice effectively each subject at a time Consist of Chemistry subject having fill in the blanks match the column mcqs one word or chemical term identification of gases state the observation define and explain the following IUPAC Nomenclature short answers numericals chemical tests figure or table based questions balancing and writing the structural formula etc Our handbook will help you study and practice well at home Why should you trust Oswal Books Oswal Publishers Oswal Publishers has been in operation since 1985 Over the past 30 years we have developed content that aids students and teachers in achieving excellence in education We create content that is extensively researched meticulously articulated and comprehensively edited catering to the various National and Regional Academic Boards in India How can you benefit from Oswal Most Likely ICSE Chemistry Question Bank for 10th Class Our handbook is strictly based on the latest syllabus prescribed by the council and is categorized chapterwise topicwise to provides in depth knowledge of different concept questions and their weightage to prepare you for Class 10th ICSE Board Examinations 2022 Having one subject per book including chapter at a glance word of advice by experts each category of our question bank covers the entire syllabus at a time Apart from study material frequently asked previous year s board questions and insightful answering tips and suggestions for students our question bank also consists of numerous tips and tools to improve study techniques for any exam paper Students can create vision boards to establish study schedules and maintain study logs to measure their progress With the help of our handbook students can also identify patterns in question types and structures allowing them to cultivate more efficient answering methods Our book can also help in providing a comprehensive overview of important topics in each subject making it easier for students to solve for the exams **SCIENCE FOR TENTH CLASS PART 2 CHEMISTRY** LAKHMIR SINGH, A series of six books for Classes IX and X according to the CBSE syllabus Each class divided into 3 parts Part 1 Physics Part 2 Chemistry Part 3 Biology *PGT Chemistry Question Bank Chapterwise - for PGT Teachers* Mocktime Publication, PGT Chemistry Question Bank Chapterwise for PGT Teachers **Science for Tenth Class Part 2 Chemistry** Lakhmir Singh & Manjit Kaur, A series of books for Classes IX and X according to the CBSE syllabus and CCE Pattern *CSIR NET Chemical Science (Chemistry) [Question Bank] Chapter Wise Question Answer of All Units 4000 +[MCQ] As Per updated Syllabus*

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combine to provide the physical basis of Life

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