

Unit 2: The Periodic Table Review Sheet

U2L1: PT History

- On the present day Periodic Table, elements are arranged in order of increasing atomic number
- Groups:** vertical columns on the periodic table
 - Elements in the same group have similar chemical properties - because they have the same number of valence electrons
 - Valence electron: outermost electrons (lost # in the electron configuration)
- Periods:** horizontal rows on the periodic table

U2L2: Parts of the Periodic Table

- Elements are classified as either metals, nonmetals, or metalloids
 - Classification can be determined by location on the periodic table or by different properties

1

H

6

C

12.01

4

14

2

4

18

He

Group 1

Group 18

Group 1

Group 2

Group 3

Group 4

Group 5

Group 6

Group 7

Group 8

Group 9

Group 10

Group 11

Group 12

Group 13

Group 14

Group 15

Group 16

Group 17

Group 18

Li

Be

Sc

Ti

V

Cr

Mn

Fe

Co

Ni

Cu

Zn

Ga

Ge

As

Se

Br

Kr

Rb

Sr

Y

Zr

Nb

Mo

Tc

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Rh

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Rn

Fr

Ra

Ac

Rf

Db

Sg

Bh

Hs

Mt

Ds

Rg

Cn

Uut

Uuq

Uup

Uuh

Uus

Uuo

Periodic Table Review Pt 2

Gary Raham



Periodic Table Review Pt 2:

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