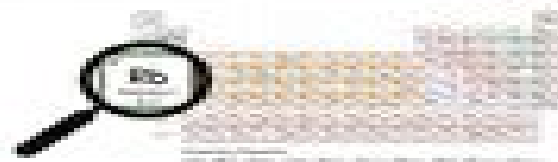


Periodic Table: Scavenger Hunt



Read the information below and answer the questions using the periodic table on page 2.

The **periodic table** is a graphical representation of all known chemical elements. Each box on the periodic table contains information about an element. The labeled image below shows the box on the periodic table for the element hydrogen.



The elements are ordered by their atomic number. Elements in the periodic table are also grouped by their chemical properties. The periodic table is color coded based on these properties.

Use the periodic table on page 2 to find the element or elements that match each description.

1. The **alkali metal** with the atomic number 19. _____
2. The **nonmetal** with an atomic mass of 32.06 _____
3. The **other metal** with an atomic mass of 26.98 _____
4. The **metalloid** with the atomic number 32 _____
5. The **noble gas** with the smallest atomic mass _____
6. The **transition metal** with the chemical symbol Au _____
7. The **lanthanide** with the atomic number 60 _____
8. The **actinide** with an atomic mass of 232.04 _____
9. The **transition metal** with the smallest atomic mass and an atomic number between 27 and 30 _____
10. The two **alkaline earth metals** with chemical symbols that start with the letter "B" a. _____ b. _____

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**Thirumalaisamy P.
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Periodic Table Scavenger Hunt Answers 8th Grade Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Periodic Table Scavenger Hunt Answers 8th Grade**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we shall delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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