

Periodic Table Named-Groups Worksheet – 2

DIRECTIONS: Use the Periodic Table shown below to answer each question. using the letters in each "box" to identify the required element(s).

The diagram shows a crossword puzzle grid with the following letters placed in specific cells:

- Row 1:** A (top left), B (top right)
- Row 2:** C (left), D (right)
- Row 3:** F (left), G (middle), H (right), K (far right)
- Row 4:** L (left), I (right)
- Row 5:** J (left), Z (left), M (middle), N (right), O (far right), P (far right)
- Row 6:** Q (left), R (left), S (middle), T (middle), U (middle)

Below the main grid, there is a separate row of cells. Two arrows originate from the letter 'R' in the main grid:

- One arrow points to the cell containing 'V'.
- Another arrow points to the cell containing 'X'.

The cells in this bottom row are arranged as follows:

- Row 7: V, (empty), (empty), (empty), (empty), (empty), W, (empty), (empty), (empty), (empty), (empty)
- Row 8: X, (empty), (empty), (empty), (empty), (empty), Y, (empty), (empty), (empty), (empty), (empty)

List elements in alphabetical order when more than 1 element answers a question.

- _____ 1) Element **A** is in a box all by itself because: **A)** it is really a Noble Gas **B)** it is a gas **C)** it is NOT a gas **D)** it does NOT share the properties of the other elements in its group **E)** it is a liquid
- _____ 2) Which of these elements is/are highly reactive? **A)** "C" **B)** "E" **C)** "K" **D)** A, B, and C **E)** A and B only
- _____ 3) What is the letter of the element whose atomic number is closest to the element labeled "E"?
- _____ 4) Which letter labels the halogen element with the lowest atomic number?
- _____ 5) Which of these elements is the least reactive? **A)** "D" **B)** "Y" **C)** "A" **D)** "O" **E)** "B"
- _____ 6) Which of these elements is/are NOT transition elements/metals (list ALL that are not transition elements)? **A)** "C" **B)** "N" **C)** "K" **D)** "N" **E)** "Y"

continued on other side

Periodic Table Packet Graphs Worksheets Answers

G Orfield



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