

Element Name, Atom or Ion?	Isotope Symbol	Atomic Number	# of protons	# of neutrons	Mass Number	# of electrons	Charge
	$^{23}_{11}\text{Na}$					10	
Nitrogen ion					15		
	$^{32}_{16}\text{S}^{2-}$					20	
Sulfur atom					33		

6. Differences between the terms:

- atom & ion
- cation & anion
- single atom ion & polyatomic ion

7. Rewrite the names, symbols and charges for all 22 polyatomic ions you have been memorizing.

8. After considering all of the types of compounds you have learned to name (listed below), label them as binary ionic compounds, binary molecular compounds, or ionic compounds.

- Type I
- Type II
- Type III
- Acids
- Compounds containing polyatomic ions

SKED Student can identify ion charges on specific elements.

9. Write the ion charge for the following symbols. Provide an additional element or group of elements that would also have that same charge:

- Hg
- F
- Ce
- Se

SKED Student can write & name formulas for Type I, II, Acids & Polyatomic ion containing compounds.

Periodic Table Mixed Review Answer Key

Bert McCarty, Philip Brown



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