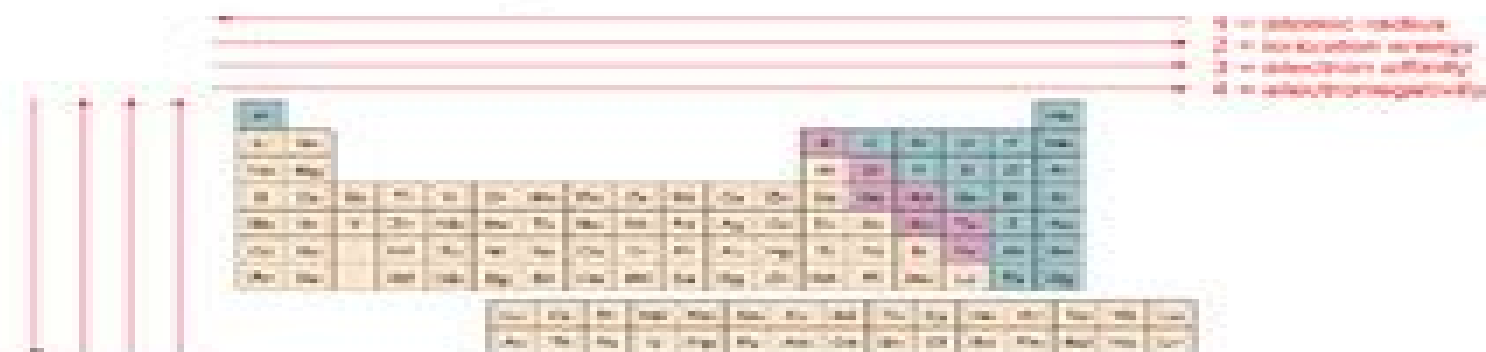


1. Match each definition with the correct trend below.

- | | |
|----------------------|--|
| d. Ionization Energy | a. The energy change when one mole of electrons is added to one mole of gaseous atoms, forming negatively charged ions. |
| c. Atomic Radius | b. A measure of an atom's ability to attract electrons in a chemical bond when forming a compound. |
| b. Electronegativity | c. The distance from the nucleus to the outermost electron in an atom, or half the distance between two nuclei in a diatomic molecule. |
| a. Electron Affinity | d. The energy required to remove one mole of electrons from one mole of gaseous atoms. |

2. Draw and label arrows on the periodic table below that indicate the direction of each trend.



3. Arrange the following elements in order of increasing atomic radius:

Be, Cl, F, I

F, Cl, Be, I

4. Rank the following elements in order of decreasing atomic radius:

C, Li, B, Be

Li, Be, B, C

5. Place the following elements in order of increasing ionization energy:

Li, K, F, O

K, Li, O, F

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Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 Holt Chemistry Ralph Thomas Myers, 2004 **Current Trends in Atomic Physics** Antoine Browaeys, Thierry Lahaye, Trey Porto, Charles S.

Adams, Matthias Weidemüller, Leticia F. Cugliandolo, 2019-05-16 This book gathers the lecture notes of courses given at Session CVII of the summer school in physics entitled Current Trends in Atomic Physics and held in July 2016 in Les Houches France Atomic physics provides a paradigm for exploring few body quantum systems with unparalleled control In recent years this ability has been applied in diverse areas including condensed matter physics high energy physics chemistry and ultra fast phenomena as well as foundational aspects of quantum physics This book addresses these topics by presenting developments and current trends via a series of tutorials and lectures presented by international leading investigators

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contributions from highly experienced invited experts

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