

## VICTORY 2026 (Class 10th)

## CHEMISTRY

## Periodic Properties and Variation of Properties

## DHA-02

- Q1** Which of the following elements has 2 shells, and both are completely filled?  
 (A) Helium (B) Neon  
 (C) Calcium (D) Boron
- Q2** Why group one elements are called alkali metals?  
 (A) When combined with water, their oxides produce acidic solutions.  
 (B) When their oxides and hydroxides are treated with water, they produce alkaline solutions.  
 (C) When their peroxides are treated with water, they produce alkaline solutions.  
 (D) When treated with water, their hydroxides produce acidic solutions.
- Q3** Which of the following gives the correct order of atomic radii of O, F and N?  
 (A) O, F, N (B) N, F, O  
 (C) O, N, F (D) F, O, N
- Q4** Where would you locate the element with electronic configuration 2,8 in the Modern Periodic Table?  
 (A) Group 8 (B) Group 2  
 (C) Group 18 (D) Group 10
- Q5** The order of increasing atomic size is:  
 (A)  $I < Br < Cl < F$   
 (B)  $I > Br > Cl > F$   
 (C)  $Br > I > Cl > F$   
 (D)  $Cl > Br > F > I$
- Q6** Newlands arranged the elements known at that time in an increasing order of their \_\_\_\_\_ masses.  
 (A) electrons  
 (B) protons  
 (C) atomic  
 (D) ions
- Q7** As per Mendeleev periodic table properties of elements are periodic function of their \_\_\_\_\_.  
 (A) Atomic number  
 (B) Atomic mass  
 (C) Molecular mass  
 (D) all of the above
- Q8** Demerits of Mendeleev periodic table are  
 (A) position of Hydrogen  
 (B) Position of Isotopes  
 (C) Position of Co and Nickel  
 (D) all of the above
- Q9** Who made modern periodic table  
 (A) Mendeleev (B) John Dalton  
 (C) John Newland (D) Henry Moseley
- Q10** The entire modern periodic table can be divided into how many blocks  
 (A) 2 blocks (B) 3 blocks  
 (C) 4 blocks (D) 5 blocks



# Periodic Variation Of Properties Answer Key

**N Noddings**



## **Periodic Variation Of Properties Answer Key:**

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**Operant-Pavlovian Interactions** Hank Davis, Harry M. B. Hurwitz, 2021-09-30 The first important distinction between operant and Pavlovian conditioning was made in 1928 by Polish scientists Konorski and Miller Unaware of their work Skinner proposed a similar analysis in 1935 of the manner in which operant and Pavlovian conditioning might differ and interact Konorski and Miller responded to Skinner s statement and by 1937 the now classic debate over two types of conditioned reflexes was in high gear In the years before publication the attention of many learning theorists had returned to the fundamental question of whether there are identifiably different forms of learning The present volume originally published in 1977 contains chapters that reassess our basic learning paradigms of the time They deal with the definitional problems of isolating operant and Pavlovian conditioning as well as the attempt to analyze the inevitable interactions that follow These issues are examined in a variety of settings some authors deal with operant Pavlovian interactions directly by devising procedures to generate them others examine operant Pavlovian interactions by examining their possible contribution to established conditioning paradigms

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