

Electron Configurations of Ions

- For **negative** ions, simply add electrons to the subshells, continuing from where the neutral atom left off.
- For **positive** ions, remove **p, s, and then d** electrons!
- **Examples:** O^{2-} , Sn^{4+}
- **Practice:** Pg. 157 #28 (a, c, e, etc.)

Small Scale Lab Electron Configurations Of Ions

Jin-Ying Zhang



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