

Spring Chapter 6

1. Consider the following energy level diagram for a reaction.

(Energy)

at reactants is 100 kJ

(kJ/mol)

at products is 200 kJ

(kJ/mol)

2. Which of the following is the correct energy diagram for the reaction?



3. A reaction has an activation energy of 100 kJ/mol. The reactants have an energy of 100 kJ/mol. What is the energy of the products?

• 100 kJ/mol
• 200 kJ/mol
• 0 kJ/mol
• 100 kJ/mol

Answer: 200 kJ/mol

4. Which of the following is the correct energy diagram for the reaction?

5. Which of the following is the correct energy diagram for the reaction?

6. Which of the following is the correct energy diagram for the reaction?

7. Which of the following is the correct energy diagram for the reaction?

8. Which of the following is the correct energy diagram for the reaction?

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Didier Musso



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