CHEMISTRY TEST

60 min Pons No calculator

Directions: Each set of lettered choices below refers to the numbered questions, formulas, or statements immediately following it. Select the one lettered choice that best answers each question or best fits each formula or statement and then blacken the corresponding space on the answer sheet. A choice may be used once, more than once, or not at all in each set.

Questions 1-4

- (A) Thermometer
- (B) Separatory funnel
- (C) Flame spectrophotometer
- (D) Buretse
- (E) Volumetric flask
- Used to study the characteristic emission lines produced by excited atoms of elements
- Used for the preparation of solutions of specific concentrations when a solid is used as the solute
- Commonly used for removing the water layer from a system containing oil and water
- 4. Used for volume measurements during titration

Questions 5-9

- CAN Linear.
- (B) Bent (V-shaped)
- (C) Trigonal, planar
- (D) Pyramidal
- (E) Regular tetrahedral

From the list above, select the shape that describes each of the following.

- 5. H,O
- 6. BI
- 7. NH.
- S. Ber.
- Q. NIH.

Questions 10-12

- (A) NaHCO₂
- (B) H,50.
- (C) CH.
- (D) Ca(OH)₂
- (E) C.H.
- 10. A significant constituent of baking powders
- 11. A constituent of gasoline
- 12. A major constituent of natural gas

Questions 13-15

- (A) $3 O_2(g) = 2 O_2(g)$
- (B) OH' + H₂O' = 2 H₂O
- (C) BaCl₂·2 H₂O(s) △ BaCl₂(s) + 2 H₂O(g)
- (D) Ca2 + CO-2 - CaCO-(a)
- (E) Fe + Cu2 = Fe2 + Cu
- 13. Represents an oxidation-reduction reaction
- Involves the formation of an ionic precipitate from a solution
- 15. Represents a Brönsted acid-base reaction

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needs Packed with proven test tips strategies and advice to help you master the test 6 full length practice SAT II Chemistry Subject tests Each test question is answered in complete detail with easy to follow easy to grasp explanations The book s handy Periodic Table of Elements allows for quick answers on the elements appearing on the exam TABLE OF CONTENTS About Research and Education Association Independent Study Schedule CHAPTER 1 ABOUT THE SAT II CHEMISTRY SUBJECT TEST About This Book About The Test How To Use This Book Format of the SAT II Chemistry Scoring the SAT II Chemistry Score Conversion Table Studying for the SAT II Chemistry Test Taking Tips CHAPTER 2 COURSE REVIEW Gases Gas Laws Gas Mixtures and Other Physical Properties of Gases Dalton s Law of Partial Pressures Avogadro s Law The Mole Concept Avogadro s Hypothesis Chemical Compounds and Formulas Mole Concept Molecular Weight and Formula Weight Equivalent Weight Chemical Composition Stoichiometry Weight and Volume Calculations Balancing Chemical Equations Calculations Based on Chemical Equations Limiting Reactant Calculations Solids Phase Diagram Phase Equilibrium Properties of Liquids Density Colligative Properties of Solutions Raoult s Law and Vapor Pressure Osmotic Pressure Solution Chemistry Concentration Units Equilibrium The Law of Mass Action Kinetics and Equilibrium Le Chatelier's Principle and Chemical Equilibrium Acid Base Equilibria Definitions of Acids and Bases Ionization of Water pH Dissociation of Weak Electrolytes Dissociation of Polyprotic Acids Buffers Hydrolysis Thermodynamics I Bond Energies Some Commonly Used Terms in Thermodynamics The First Law of Thermodynamics Enthalpy Hess's Law of Heat Summation Standard States Heat of Vaporization and Heat of Fusion Thermodynamics II Entropy The Second Law of Thermodynamics Standard Entropies and Free Energies Electrochemistry Oxidation and Reduction Electrolytic Cells Non Standard State Cell Potentials Atomic Theory Atomic Weight Types of Bonds Periodic Trends Electronegativity Quantum Chemistry Basic Electron Charges Components of Atomic Structure The Wave Mechanical Model Subshells and Electron Configuration Double and Triple Bonds Organic Chemistry Nomenclature and Structure Alkanes Alkenes Dienes Alkynes Alkyl Halides Cyclic Hydrocarbons Aromatic Hydrocarbons Aryl Halides Ethers and Epoxides Alcohols and Glycols Carboxylic Acids Carboxylic Acid Derivatives Esters Amides Arenes Aldehydes and Ketones Amines Phenols and Quinones Structural Isomerism SIX PRACTICE EXAMS Practice Test 1 Answer Key Detailed Explanations of Answers Practice Test 2 Answer Key Detailed Explanations of Answers Practice Test 3 Answer Key Detailed Explanations of Answers Practice Test 4 Answer Key Detailed Explanations of Answers Practice Test 5 Answer Key Detailed Explanations of Answers Practice Test 6 Answer Key Detailed Explanations of Answers THE PERIODIC TABLE EXCERPT About Research OR 2 To demonstrate proficiency in Chemistry The SAT II Chemistry exam is designed for students who have taken one year of college preparatory chemistry Who Administers The Test The SAT II Chemistry Subject Test is developed by the College Board and administered by Educational Testing Service ETS The test development process involves the assistance of educators throughout the country and is designed and implemented to ensure that the content and difficulty level of the test are appropriate When Should the SAT II Chemistry be Taken If you are applying to a college that requires Subject Test scores as part of the admissions process you should take the SAT II Chemistry Subject Test toward the end of your junior year or at the beginning of your senior year If your scores are being used only for placement purposes you may be able to take the test in the spring of your senior year For more information be sure to contact the colleges to which you are applying When and Where is the Test Given The SAT II Chemistry Subject Test is administered five times a year at many locations throughout the country mostly high schools To receive information on upcoming administrations of the exam consult the publication Taking the SAT II Subject Tests which may be obtained from your guidance counselor or by contacting College Board SAT Program P O Box 6200 Princeton NJ 08541 6200 Phone 609 771 7600 Website http www collegeboard com Is There a Registration Fee Yes There is a registration fee to take the SAT II Chemistry Consult the publication Taking the SAT II Subject Tests for information on the fee structure Financial assistance may be granted in certain situations To find out if you qualify and to register for assistance contact your academic advisor HOW TO USE THIS BOOK What Do I Study First Remember that the SAT II Chemistry Subject Test is designed to test knowledge that has been acquired throughout your education Therefore the best way to prepare for the exam is to refresh yourself by thoroughly studying our review material and taking the sample tests provided in this book They will familiarize you with the types of questions directions and format of the SAT II Chemistry Subject Test To begin your studies read over the review and the suggestions for test taking take one of the practice tests to determine your area s of weakness and then restudy the review material focusing on your specific problem areas The course review includes the information you need to know when taking the exam Be sure to take the remaining practice tests to further test yourself and become familiar with the format of the SAT II Chemistry Subject Test When Should I Start Studying It is never too early to start studying for the SAT II Chemistry test The earlier you begin the more time you will have to sharpen your skills Do not procrastinate Cramming is not an effective way to study since it does not allow you the time needed to learn the test material The sooner you learn the format of the exam the more comfortable

you will be when you take the exam FORMAT OF THE SAT II CHEMISTRY The SAT II Chemistry is a one hour exam consisting of 85 multiple choice questions The first part of the exam consists of classification questions This question type presents a list of statements or questions that you must match up with a group of choices lettered A through E Each choice may be used once more than once or not at all The exam then shifts to relationship analysis questions which you will answer in a specially numbered section of your answer sheet You will have to determine if each of two statements is true or false and if the second statement is a correct explanation of the first The last section is composed strictly of multiple choice questions with choices lettered A through E Material Tested The following chart summarizes the distribution of topics covered on the SAT II Chemistry Subject Test Topic Percentage Number of Questions Atomic however do not review too much at any one time Concentrate on one problem area at a time by reviewing the questions and explanations and by studying our review until you are confident you completely understand the material Keep track of your scores By doing so you will be able to gauge your progress and discover general weaknesses in particular sections You should carefully study the reviews that cover your areas of difficulty as this will build your skills in those areas TEST TAKING TIPS Although you may be unfamiliar with standardized tests such as the SAT II Chemistry Subject Test there are many ways to acquaint yourself with this type of examination and help alleviate your test taking anxieties Become comfortable with the format of the exam When you are practicing to take the SAT II Chemistry Subject Test simulate the conditions under which you will be taking the actual test Stay calm and pace yourself After simulating the test only a couple of times you will boost your chances of doing well and you will be able to sit down for the actual exam with much more confidence Know the directions and format for each section of the test Familiarizing yourself with the directions and format of the exam will not only save you time but will also ensure that you are familiar enough with the SAT II Chemistry Subject Test to avoid nervousness and the mistakes caused by being nervous Do your scratchwork in the margins of the test booklet You will not be given scrap paper during the exam and you may not perform scratchwork on your answer sheet Space is provided in your test booklet to do any necessary work or draw diagrams If you are unsure of an answer guess However if you do guess guess wisely Use the process of elimination by going through each answer to a question and ruling out as many of the answer choices as possible By eliminating three answer choices you give yourself a fifty fifty chance of answering correctly since there will only be two choices left from which to make your guess Mark your answers in the appropriate spaces on the answer sheet Fill in the oval that corresponds to your answer darkly completely and neatly You can change your answer but remember to completely erase your old answer Any stray lines or unnecessary marks may cause the machine to score your answer incorrectly When you have finished working on a section you may want to go back and check to make sure your answers correspond to the correct questions Marking one answer in the wrong space will throw off the rest of your test whether it is graded by machine or by hand You don't have to answer every question You are not penalized if you do not answer every question The only penalty results from answering a

question incorrectly Try to use the guessing strategy but if you are truly stumped by a question remember that you do not have to answer it Work quickly and steadily You have a limited amount of time to work on each section so you need to work quickly and steadily Avoid focusing on one problem for too long Before the Test Make sure you know where your test center is well in advance of your test day so you do not get lost on the day of the test On the night before the test gather together the materials you will need the next day Your admission ticket Two forms of identification e q driver s license student identification card or current alien registration card Two No 2 pencils with erasers Directions to the test center A watch if you wish but not one that makes noise as it may disturb other test takers. On the day of the test you should wake up early after a good night's rest and have breakfast Dress comfortably so that you are not distracted by being too hot or too cold while taking the test Also plan to arrive at the test center early This will allow you to collect your thoughts and relax before the test and will also spare you the stress of being late If you arrive after the test begins you will not be admitted to the test center and you will not receive a refund During the Test When you arrive at the test center try to find a seat where you feel most comfortable Follow all the rules and instructions given by the test supervisor If you do not you risk being dismissed from the test and having your scores canceled Once all the test materials are passed out the test instructor will give you directions for filling out your answer sheet Fill this sheet out carefully since this information will appear on your score report After the Test When you have completed the SAT II Chemistry Subject Test you may hand in your test materials and leave Then go home and relax When Will I Receive My Score Report and What Will It Look Like You should receive your score report about five weeks after you take the test This report will include your scores percentile ranks and interpretive information **Cracking the SAT II.** Theodore Silver, Princeton Review (Firm), 2001-03-15 The Princeton Review realizes that acing the SAT II Chemistry exam is very different from getting straight As in school They don't try to teach students everything there is to know about chemistry only what they ll need to score higher on the exam There s a big difference InCracking the SAT II Chemistry The Princeton Review will teach test takers how to think like the test makers and Learn test taking strategies that will help students outsmart the test and improve scores Ace the exam by becoming familiar with the format Use the Process of Elimination and the divide and conquer method to solve complicated problems Perfect test taking skills with practice questions and detailed answer explanations This book includes 2 full length simulated SAT II Chemistry exams All of the sample test questions are just like the ones test takers will see on the actual exam and every solution is fully explained Contents Include I Introduction II Test Strategies III Some Basic Stuff Mass Volume Density Pressure Energy Temperature and Specific Heat IV Elements Atoms and Ions Atoms and Elements V Chemical Reaction and Stoichiometry Molecules The Mole Chemical Reactions Reaction Stoichiometry Entropy Enthalpy Spontaneity and Gibbs Free Energy VI Electron configurations and Radioactivity Electrons and Orbitals Radioactivity VII The Periodic Table and Bonding The Periodic Table More About the Periodic Table Some Important Trends VIII Solids Liquids and Gases Gases Intermolecular

Forces Phase Changes Energy and Phase Changes IX Solutions Solutions Concentrations Solubility and Saturation X Kinetics and Equilibrium Kinetics Factors that Affect Reaction Rate Reversible Reactions and Chemical Equilibrium Le Chatelier s Principle XI Acids and Bases Acids and Bases Titration XII Redox and Electrochemistry Oxidation and Reduction Electrochemistry XIII Organic Chemistry Hydrocarbons Functional Groups XIV Laboratory Safety Rules Accuracy Significant Figures Lab Procedures Laboratory Equipment XV Practice Tests Cracking the SAT II Chemistry Theodore Silver, Princeton Review (Firm), 2005-03 Why The Princeton Review 1 We Know the SAT Chemistry Subject Test The experts at The Princeton Review have spent many years researching the SAT Chemistry Subject Test as well as numerous other standardized tests We re confident this guide delivers the most current and complete information you need to ace this test 2 We Get Results Our inventive approach to standardized test taking has revolutionized the test prep industry and made our courses and tutoring for the SAT and SAT Subject Tests the most popular anywhere The same proven techniques we teach in our courses are also covered in this book 3 We Understand Students Each year we help more than two million students score higher on standardized tests and gain admission to top schools with our books courses tutors and online tools 4 And If It s on the SAT Chemistry Subject Test It's in This Book The Princeton Review realizes that acing the SAT Chemistry Subject Test is very different from getting straight A s in school We don t try to teach you everything there is to know about chemistry only the techniques and information you ll need to maximize your score In Cracking the SAT Chemistry Subject Test we ll teach you how to think like the test writers and Master test taking strategies that will improve your score Ace the exam by familiarizing yourself with its format Use Process of Elimination and other proven test taking techniques to solve complicated problems Perfect your test taking skills with practice guestions and detailed answers and explanations This book includes three full length practice SAT Chemistry Subject Tests All of our practice test questions are just like those you ll see on the actual test and we fully explain every question Attend Free Practice Tests and Strategy Sessions We re not just good on paper you should see us live The Princeton Review frequently offers free events to students and parents Evaluate Your Options Thousands of students prepare for standardized tests with our books courses and tutoring programs Get on the Inside Track for College Admissions Gaining admission to top colleges takes more than a high test score Other important qualifiers may include a strong admissions essay GPA and volunteer work To learn more about our many books programs and services go to PrincetonReview com or call us at 800 2Review McGraw-Hill's SAT Subject Test: Chemistry, 2ed Thomas Evangelist, 2009-01-22 Expert guidance on the Chemistry SAT Subject Test Many colleges and universities require you to take one or more SAT II Subject Tests to demonstrate your mastery of specific high school subjects McGraw Hill's SAT Subject Test Chemistry is written by experts in the field and gives you the guidance you need perform at your best This book includes Full length sample tests Diagnostic exams to help students pinpoint strengths and weaknesses Hundreds of questions with complete explanations Step by step reviews of all topics covered in each exam Strategies to help students get

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find your weakness and study the topic specifically You need have your textbook on hand and study the topic as necessary SAT Chemistry Test Prep Review--Exambusters Flash Cards SAT II Exambusters, 2017-12-01 SAT CHEMISTRY Study Guide 700 questions and answers Essential definitions formulas concepts and sample problems Topics Introduction Matter Atoms Formulas Moles Reactions Elements Periodic Table Electrons Chemical Bonds Heat Gases Phase Changes Solutions Reaction Rates Equilibrium Acids and Bases Oxidation and Reduction Introduction to Organic Chemistry Radioactivity EXAMBUSTERS SAT II Prep Workbooks provide comprehensive SAT II review one fact at a time to prepare students to take practice SAT II tests Each SAT II study guide focuses on fundamental concepts and definitions a basic overview to begin studying for the SAT II exam Up to 600 questions and answers each volume in the SAT II series is a quick and easy focused read Reviewing SAT II flash cards is the first step toward more confident SAT II preparation and ultimately Cracking the SAT II Biology E/M Subject Test Princeton Review, Judene Wright, 2005 higher SAT II exam scores Reviews the key concepts of biology and includes two full length practice tests **How to Prepare for the SAT II Chemistry** Joseph A. Mascetta, 2002-09-01 Up to date preparation for the SAT II Chemistry includes a diagnostic test and four full length practice tests to help students assess their strengths and weaknesses All test questions are answered and explained Subject review chapters cover all test topics structure of matter states of matter solutions reaction types stoichemistry equilibrium and reaction rates thermodynamics descriptive chemistry and much more the SAT Subject Test in Chemistry Joseph A. Mascetta, 2006-07-01 Up to date preparation for the SAT Subject Test in Chemistry formerly called the SAT II Chemistry includes a diagnostic test and four full length practice tests All tests are presented with answers and explanations Subject review chapters with additional questions and answers cover all test topics structure of matter states of matter solutions reaction types stoichoimetry equilibrium and reaction rates thermodynamics descriptive chemistry and much more New in this edition are explained answers for all end of chapter practice questions as well as chapter objectives that guide students on what they should look for when they review each chapter USMLE Step 2 Rose S. Fife, Irene E. Chen, Craig L. Levitz, James K. Min, Gopi Rana-Mukkavilli, T. M. Worner, 1998 Prepare for the USMLE Step 2 Exam with three full length practice exams detailed explanations to every exam question and a coaching review Over 2 100 practice questions cover topics in internal medicine obstetrics pediatrics preventive medicine surgery and more testing the student's ability to practice medicine under supervision An essential study guide for those seeking a top score on this **Cracking the CBEST, 2nd Edition** Rick Sliter, 2002 The CBEST is required for teacher certification in important exam California and Oregon Each new teacher must receive certification which means that they must pass the CBEST This book shows them how Canadian History: Pre-Colonization to 1867 Essentials Terry Crowley, 2013-01-01 REA s Essentials provide quick and easy access to critical information in a variety of different fields ranging from the most basic to the most advanced As its name implies these concise comprehensive study guides summarize the essentials of the field covered

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