

Lab 6

SATURATION AND ATMOSPHERIC STABILITY

Materials Needed

- calculator
- ruler

Introduction

How do clouds form? The concept of saturation, introduced in the previous lab, is important here. The two basic means by which air achieves saturation (relative humidity = 100%) are cooling and increasing water vapor content. In this lab we will examine several common atmospheric processes that will lead to either or both of these phenomena. We will focus particularly on how air changes as it rises from the earth's surface and how these changes relate to atmospheric stability.

Mixing

Saturation occasionally occurs when two air masses mix. Mixing usually involves a change in both water vapor content and temperature. Consider, for example, two unsaturated air samples, A and B:

Air Sample	A	B
Temperature	10°C	38°C
Relative humidity	75%	75%

1. Use Table 6-1 to find the saturation mixing ratio (m_s) of air samples A and B.

$$A: m_s = \underline{\hspace{2cm}} \text{ g kg}^{-1}$$

$$B: m_s = \underline{\hspace{2cm}} \text{ g kg}^{-1}$$

2. What are the mixing ratio (m) values of samples A and B?

$$A: m = \underline{\hspace{2cm}} \text{ g kg}^{-1}$$

$$B: m = \underline{\hspace{2cm}} \text{ g kg}^{-1}$$

3. Use the temperature and mixing ratio data to plot the two air samples in Figure 6-1.

Assume that equal parts of these two air masses mix together. The resulting temperature is halfway between 10°C and 38°C, and the mixing ratio is halfway between the values that you found for the two samples.

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Jessica J Manson

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