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assertion on page 60 that for typical thermodynamic processes in which the elevation of the center of mass changes it is usually a good approximation to set w equal to w_{lab}

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temperature remains constant $\Delta t = 0$ since internal energy depends on the temperature $\Delta u = 0$ from first law of thermodynamics

$\Delta u = \Delta q = \Delta w$ since $\Delta u = 0$ $\Delta q = \Delta w$ also p, v nrt as t is constant p, v