

# > Workbook answers

## Chapter 1

### Exercise 1.1

An atom contains a dense nucleus surrounded by shells of electrons. The nucleus contains the nucleons (protons and neutrons). Protons are positively charged, electrons have a negative charge and neutrons are uncharged. The protons and neutrons have the same relative mass. The mass of an electron is negligible.

### Exercise 1.2

1 and C; 2 and B; 3 and D; 4 and A

### Exercise 1.3

a A 36, B 49, C 66, D 24, E 24, F 26 b i

There is a mixture of isotopes present.

(Relative atomic mass is the weighted mean of these isotopes.) ii 52

iii There are two more protons than electrons in the strontium ion, c i 13 protons, 13 electrons, 14 neutrons ii 55 protons, 54 electrons, 78 neutrons iii 8 protons, 10 electrons, 9 neutrons

### Exercise 1.4

- a  $2^4\text{He}^{2+}$
- b Most of the atom is empty space because the nucleus is extremely small. The electrons have a much smaller mass than the alpha-particles so do not cause a change in their momentum if they do collide.
- c The positive charge on the nucleus (protons) repels the positive charge on the alpha-particle.

- d The nucleus is extremely small so very few particles get close enough to it to be repelled completely.
- e The neutrons would go straight through the foil because they are uncharged. They are not deflected by the positive charge in the nucleus.
- f Atoms of the isotope with more neutrons have a greater mass. So since density = mass/volume, the density of heavier isotopes is greater.

## Exam-style questions

### Question 1

- a There are 3 protons, which are positively charged. [1]  
There are 4 neutrons, which have no charge. [1]  
There are 3 electrons, which are negatively charged. [1]  
Protons and neutron have a relative mass of 1. [1]  
Electrons have hardly any mass / mass of about 1/2000 that of a proton. [1] b The electron configuration is the same / there is the same number of outer shell electrons. [1]  
c  $73\text{Li}^+$  (1 mark for 7 and 3 in correct places, 1 mark for  $\text{Li}^+$ ) [2]  
d It has 2 electrons and 3 protons / it has one more proton than electrons. [1]

[Total: 9]

### Question 2

- a Nickel, because it has one more proton than cobalt / nickel has 28 protons and cobalt has 27 [1]  
b Relative atomic mass depends on the weighted mean [1]

# Princeton Hall Chemistry Workbook Answers

**Camilla Rothe**



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