

**END TERM EXAMINATION**

FIRST SEMESTER [B.TECH.] DECEMBER 2016

Paper Code: ETCH-113

Subject: Applied Chemistry

Time: 3 Hours

Maximum Marks: 75

Note: Attempt any five questions including Q no.1 which is compulsory.  
Select one question from each unit.

- Q1
- (a) Define HCV and GCV. (2.5)
  - (b) Define order and molecularity of a reaction with suitable example. (2.5)
  - (c) What is biodiesel? (2.5)
  - (d) State Gibb's phase rule. (2.5)
  - (e) What is Pilling-Bedworth rule? (2.5)
  - (f) What is an eutectic mixture? (2.5)
  - (g) Explain what is happening at the triple point of water. How many phases are there in the system at this point? (2.5)
  - (h) What is synthetic petrol? (2.5)
  - (i) Name the catalyst used for the following reactions/processes:- (5)
    - (a) Fischer Tropach process
    - (b) Contact Process
    - (c) Zeigler Natta Polymerisation
    - (d) Haber's Process
    - (e) Homogeneous hydrogenation reaction

**UNIT-I**

- Q2
- (a) Differentiate the following:- (2.5x4=10)
    - (i) Chemical fuel and nuclear fuel
    - (ii) High temperature and low temperature carbonization
    - (iii) Fixed and Fluidized bed catalytic cracking.
    - (iv) Octane and octane number
  - (b) What is sweetening of petrol? (2.5)

- Q3
- (a) Explain how the calorific value of a fuel can be determined by Bomb's calorimeter with the help of a neat diagram. (6.5)
  - (b) Calculate GCV and NCV of a gaseous fuel from the following data obtained from calorimeter experiment. (6)
    - Volume of gaseous fuel burnt at STP = 0.08m<sup>3</sup>
    - Temperature of inlet water = 25°C
    - Temperature of outlet water = 32°C
    - Weight of water produced by steam Condensation = 0.01 kg.
    - Latent heat of steam = 587 Kcal/kg

**UNIT-II**

- Q4
- (a) What is "degree of freedom" of a system? Calculate degrees of freedom of the following systems:- (10)
    - (i) Water (liquid)  $\leftrightarrow$  water vapour at 1 atm.
    - (ii)  $S_r \leftrightarrow S_m \leftrightarrow S_l \leftrightarrow S_v$
    - (iii) An aqueous solution of glucose
    - (iv) Pure crystals of  $CuSO_4 \cdot 5H_2O$
  - (b) What are cooling curves? (2.5)
- Q5
- (a) Explain Langmuir Hinshelwood mechanism of catalysis in detail. (6)
  - (b) How does a catalyst increase the rate of reaction by affecting the reaction pathway? Will it affect the rate of backward reaction in case of a reversible reaction? Also describe auto catalysis with suitable examples. (6.5)

P.T.O.

ETCH-113

R/2

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**Good Schools Guide** Lucas Publications,2003-06-02

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**Models of Computation in Context** Benedikt Löwe,Dag Normann,Ivan Soskov,Alexandra Soskova,2011-06-22 This book constitutes the refereed proceedings of the 7th Conference on Computability in Europe CiE 2011 held in Sofia Bulgaria in June July 2011 The 22 revised papers presented together with 11 invited lectures were carefully reviewed and selected with an acceptance rate of under 40% The papers cover the topics computability in analysis algebra and geometry classical computability theory natural computing relations between the physical world and formal models of computability theory of transfinite computations and computational linguistics

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The Transformation of the 1966 Soviet Input-output Table from Producers to Adjusted Factor Cost Values Steven Rosefielde,1975

**Web and Big Data** Xiangyu Song,Ruyi Feng,Yunliang Chen,Jianxin Li,Geyong Min,2024-05-11 The 4 volume set LNCS 14331 14332 14333 and 14334 constitutes the refereed proceedings of the 7th International Joint Conference APWeb WAIM 2023 which took place in Wuhan China in October 2023 The total of 138 papers included in the proceedings were carefully reviewed and selected from 434 submissions They focus on innovative ideas original research findings case study results and experienced insights in the areas of the World Wide Web and big data covering Web technologies database systems information management software engineering knowledge graph recommend system and big data

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