

# Physical Science – 2nd Semester – Final Exam Study Guide – (2012–2013)

## 1. Physical Science Basics

- What tool(s) would you want to use to find the:
  - Mass of an object? triple beam balance Basic SI Unit? kg
  - Volume of an object? ruler or water displacement method Basic SI Unit? cm<sup>3</sup> or ml
  - Length of an object? Ruler Basic SI Unit? meter
  - Density of an object? graduated cylinder Basic SI Unit? ml
- What is the formula for calculating density? mass/volume
  - What is the density of a cube that is 25 g and has a volume of 3 cm<sup>3</sup>? 8g/cm<sup>3</sup>
  - Compare the density of solid H<sub>2</sub>O to liquid H<sub>2</sub>O, and explain what observable effect this has. solid H<sub>2</sub>O less dense than liquid H<sub>2</sub>O

## 2. Chemistry Basics

- Element (def.) substance made up of one kind of atom
- Atom (def.) basic building block of matter
- Compound (def.) a molecule that contains atoms or molecules of two or more different elements
- Molecule (def.) a combination of two or more of the same or different atoms that are bonded together

## 3. Model of the Atomic Evolution of the Atomic Theory

Atoms are so small that many scientists throughout history have created models to describe them.

Why have atomic theories changed over time? Better technology came around.

Scientist	Year	Major Discovery/ Contribution	Name of atomic model
John Dalton	1808	Each element is made up of small, indivisible particles called atoms. Different elements have atoms of different masses.	Billiard Ball Model
J.J. Thomson	1897	Negatively charged electron	Plum Pudding Model
Niels Bohr	1913	Electrons move <u>fixed orbits</u> around the atom's nucleus in specific layers, or <u>electron shells</u> . Atoms absorb or give off energy when the electrons move from one shell ( <u>energy level</u> ) to another.	Bohr Model
Ernest Rutherford	1909	1909 /Based on idea that positive reflects positive/ atom mostly empty space; electrons orbit randomly around positively charged nucleus	Planetary Model

## 4. Periodic Table

- The current periodic table is arranged by increasing atomic number.
- Reading the Periodic Table
  - An element's properties can be predicted from its location in the periodic table.
  - The atomic number represents the number of protons.
  - The atomic mass represents the number of protons and neutrons.

# Physical Science Final Exam Study Guide Reviews

**Robert R. Redfield**



## **Physical Science Final Exam Study Guide Reviews:**

**X-kit Exam 2004 Physical Science** ,2005      Master the GED: The Language Arts, Writing Test Peterson's,2012-06-29 Peterson's Master the GED The Language Arts Writing Test comprises three review chapters exclusively focused on your preparation for the two parts of the Writing test The material introduces each question type you will encounter in Part I and give you tips and strategies for answering each one This eBook includes an in depth writing review to help you with the sentence structure problems that make up about a third of the questions in Part I The final review chapter pays special attention to the essay you will need to write reviewing successful writing strategies and discussing the strengths and weaknesses of a number of included sample essays It concludes with strategies on how to evaluate your own practice essays

**Master the GED: The Mathematics Test** Peterson's,2012-06-29 Peterson's Master the GED The Mathematics Test will thoroughly prepare you for both parts of the GED Math Test After a brief tutorial about how to use the calculator you are allowed to use during Part I and giving you strategies for solving math problems this eBook offers in depth math reviews of numbers order and laws of operations integers fractions ratios and square roots among others algebra and descriptive statistics and geometry You will find sample questions with answer explanations throughout the review chapters for further practice

**Chemical News and Journal of Physical Science** William Crookes,James H. Gardiner,Gerald Druce,H. W. Blood-Ryan,1900      **Air University Review** ,1969      Master the GED: Practice Test 3 Peterson's,2011-12-01 Master the GED is a comprehensive guide that offers the essential test prep and review material for the high school equivalency diploma test including practice tests subject review and expert tips on how to score high on each GED test Readers will find the GED information they need to know scoring and passing requirements how to prepare and what to expect on test day

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*Nursing School Entrance Exams Prep* Kaplan Nursing, 2024-02-06 Now with a new easy to read page design Kaplan's Nursing School Entrance Exams Prep is a focused review of the HESI A2 and the Kaplan Nursing Admission Test two major nursing school entrance assessments Exam specific practice concise content review and proven test taking strategies will prepare you to face the first test of your nursing career with confidence The Best Review Four sample practice tests two for HESI A2 two for the Kaplan exam Diagnostic test to identify the topics where you need the most review Test specific icons showing which content to review for the Kaplan vs the HESI Science chapters broken out by topic anatomy physiology biology organ systems and chemistry Grammar and writing sections specifically geared to the Kaplan test Quick reference resources with frequently used math formulas and commonly misspelled words to remember Expert Guidance Kaplan's expert nursing faculty reviews and updates content regularly Practical advice for the career change nursing student We invented test prep Kaplan www.kaptest.com has been helping students for 80 years and our proven strategies have helped legions of students achieve their dreams **Physics Handbook & Study Guide Grade 11-12 IEB** Kevin Smith, 2024-02-01 A comprehensive summary of Grade 11 12 Physics Simple logical summaries with example exam questions and work through solutions The book covers the fundamentals of Grade 11 12 Physics and complements the material in any class text

CliffsNotes Praxis II: Middle School Science (0439) Glen Moulton, 2013-05-21 Your complete guide to a higher score on Praxis II Middle School Science The Praxis II Middle School Science 0439 exam is designed to measure the knowledge and competencies necessary for a beginning teacher of middle school science The 2 hour Praxis II Middle School Science 0439 exam consists of three constructed response essays and 90 multiple choice questions divided into the following content categories scientific methodology basic principles of science physical sciences life sciences earth space sciences and science technology society In CliffsNotes Praxis II Middle School Science two practice tests with complete answers and explanations help you pinpoint areas for further study while reviews and exercises address all of the test topics you'll encounter on exam day Plus proven test taking strategies help you score higher Two full length practice tests Subject reviews of every topic covered on the test Practice questions for every subject review If you're an aspiring teacher looking to take the Praxis II Middle School Science exam CliffsNotes is your ticket to scoring high at exam time **The Cambridge Review**, 1883

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