

FUNCTIONAL GROUPS OF ORGANIC COMPOUNDS

Homologous series	Structure of functional group	Example of a compound	
		Name	Structural formula
alkanes	Only C-H and C-C single bonds	ethane	$\begin{array}{c} \text{H} & & \text{H} \\ & & \\ \text{H}-\text{C} & - & \text{C}-\text{H} \\ & & \\ \text{H} & & \text{H} \end{array}$
alkenes	$\begin{array}{c} \diagup & & \diagdown \\ & \text{C}=\text{C} & \\ \diagdown & & \diagup \end{array}$	ethene	$\begin{array}{c} \text{H} & & \text{H} \\ & \diagdown & \diagup \\ & \text{C}=\text{C} & \\ & \diagup & \diagdown \\ \text{H} & & \text{H} \end{array}$
alkynes	$-\text{C}\equiv\text{C}-$	ethyne	$\text{H}-\text{C}\equiv\text{C}-\text{H}$
haloalkanes (alkyl halides)	$\begin{array}{c} \\ -\text{C}-\text{X} \\ \\ \text{(X = F, Cl, Br, I)} \end{array}$	bromoethane	$\begin{array}{c} \text{H} & & \text{H} \\ & & \\ \text{H}-\text{C} & - & \text{C}-\text{Br} \\ & & \\ \text{H} & & \text{H} \end{array}$
alcohols (alkanols)	$\begin{array}{c} \\ -\text{C}-\text{O}-\text{H} \\ \end{array}$	ethanol	$\begin{array}{c} \text{H} & & \text{H} \\ & & \\ \text{H}-\text{C} & - & \text{C}-\text{O}-\text{H} \\ & & \\ \text{H} & & \text{H} \end{array}$
aldehydes	$\begin{array}{c} \text{O} \\ \\ -\text{C}-\text{H} \end{array}$	ethanal	$\begin{array}{c} \text{H} & & \text{O} \\ & & \\ \text{H}-\text{C} & - & \text{C}-\text{H} \\ \\ \text{H} \end{array}$
ketones	$\begin{array}{c} \text{O} \\ \\ -\text{C}- \\ \end{array}$	propan-2-one	$\begin{array}{c} \text{H} & & \text{O} & & \text{H} \\ & & & & \\ \text{H}-\text{C} & - & \text{C} & - & \text{C}-\text{H} \\ & & & & \\ \text{H} & & & & \text{H} \end{array}$
carboxylic acids	$\begin{array}{c} \text{O} \\ \\ -\text{C}-\text{O}-\text{H} \end{array}$	ethanoic acid	$\begin{array}{c} \text{H} & & \text{O} \\ & & \\ \text{H}-\text{C} & - & \text{C}-\text{O}-\text{H} \\ \\ \text{H} \end{array}$
esters	$\begin{array}{c} \text{O} \\ \\ -\text{C}-\text{O}-\text{C}- \\ \end{array}$	methyl ethanoate	$\begin{array}{c} \text{H} & & \text{O} & & \text{H} \\ & & & & \\ \text{H}-\text{C} & - & \text{C}-\text{O}- & \text{C}-\text{H} \\ & & & & \\ \text{H} & & & & \text{H} \end{array}$

NAMING OF ORGANIC COMPOUNDS

The name of every organic molecule has three parts:

- The **parent name** indicates the number of C atoms in the longest carbon chain in the molecule.
- The **suffix** indicates what functional group is present.
- The **prefix** reveals the identity, location and number of substituents attached to the carbon chain.

prefix

What and where are the substituents?

parent

How many carbons?

suffix

What is the functional group/homologous series?

di - 2

tri - 3

tetra - 4

Number of carbon atoms	Parent name	Number of carbon atoms	Parent name
1	meth	5	pent
2	eth	6	hex
3	prop	7	hept
4	but	8	oct

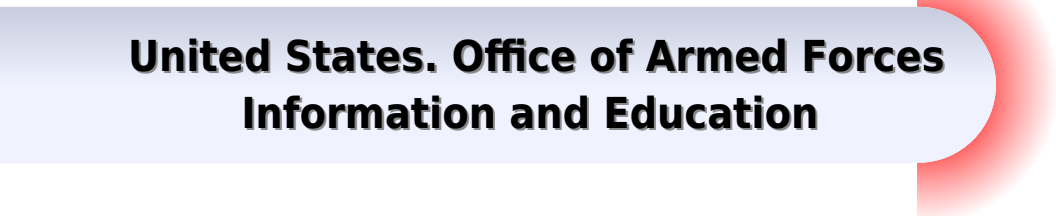


prefix, suffix and parent!

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Tetra-chloro-metha

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