

Physical Properties of Water

Water is the only natural substance that is found in all three states: solid (ice), liquid, and gas (steam). It boils at 212°F or 100°C and freezes at 32°F or 0°C. Water has a high specific heat index.

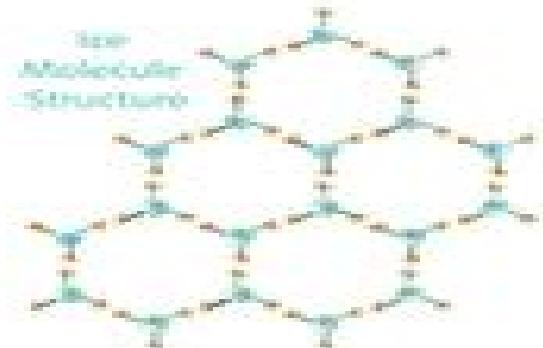
Specific heat is the amount of energy required to change the temperature of a substance. Water can absorb a lot of heat before it begins to get hot, which is why it is used at the beach; always feels better than the ocean water. Land absorbs heat faster than water.

Water's high specific heat index makes it valuable, especially to manufacturing companies. It is used as a coolant because it can absorb a lot of heat before it begins to get hot. Water can move over heated elements in a factory, absorbing the heat from machines. In a car, water is a coolant in the radiator that absorbs heat put out by the engine.

When water molecules make a change in phase or state, their molecules rearrange themselves into different patterns. In the liquid phase, water molecules are close together with no regular arrangement. The particles of a liquid vibrate, move about, and slide past each other. A solid has tightly packed particles that are usually in a regular pattern. There is very little vibration or movement of the particles of a solid. In the gaseous phase, water molecules are widely spaced apart and very loosely with no regular arrangement. Generally, at the temperature mark, matter moves to a more active state. This movement of molecules is called kinetic energy.

Most substances are the densest in their solid form. However, water is different. Solid water, or ice, floats on top of liquid water. Why is this? In order to float on water, a substance must be less dense than water. When ice is formed, the water molecules are tightly packed together, preventing them from changing shape. As for a regular pattern with the molecules held tightly apart by their bonds. This causes ice to form a crystalline lattice structure. These crystals have many open spaces throughout their structure, making ice less dense than liquid water.

Vocabulary	
specific heat	constant
kinetic energy	



	Gas	Liquid	Solid
particle arrangement	well separated — no regular arrangement	close together — no regular arrangement	tightly packed — no regular pattern
shape	assumes the shape of its container - no shape	assumes the shape of its container - no shape	retains a fixed volume and shape
kinetic energy — movement	very active	slide past, push other	rigid
fluid	fluid — flows easily	fluid — flows easily	not a fluid — does not flow easily
volume	volume of container	fixed volume	fixed volume

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Daniela Niemeyer

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