#### To Begin

You made models of atoms, elements, molecules and compounds. You discussed homeogenous and heterogenous mixtures. You have identified examples of pure substances and of mixtures.

In this activity you will work to complete sketches of increasingly microscopic views of particles. You have an opportunity to define terms and add sketches to your comp book.

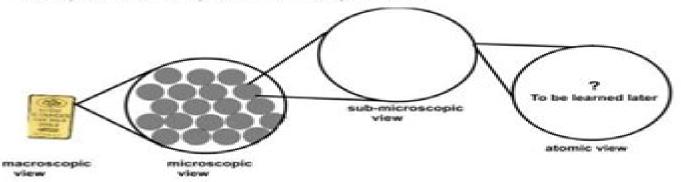
#### Learning Goals for this activity

Students will identify and/or sketch each of the following: element, compound, molecule, atom, pure substance, mixture, homogeneous and heterogeneous.

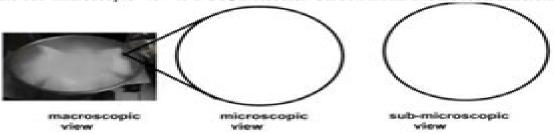
#### Process & Procedure

Discuss the example presented by your teacher with your team. Complete this worksheet on your own...only seek help from your team if you need it.

 This substance is gold. Complete the sub-microscopice view of gold. Sub-microscopic views look at the atomic make-up of the individual particles that make up a material.



- From the example of gold (above), explain the difference between the element gold and a gold atom.
- The substance is carbon dioxide (gas), CO<sub>2</sub>. Complete the microscopic and sub-microscopic views of CO<sub>2</sub>.
  Connect the sub-microscopic view to a CO<sub>2</sub> molecule. Carbon atoms should look different than oxygen atoms.



4. The

# **Physical Science 9th Grade Study Guide**

Zhe Xu

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