

QUESTION 1

- 1.1. C ✓✓
- 1.2. D ✓✓
- 1.3. B ✓✓
- 1.4. A ✓✓
- 1.5. A ✓✓

[10]

QUESTION 2

- 2.1.
- 2.1.1. A pure substance is a substance that cannot be separated into simpler components by physical methods. ✓✓
- 2.1.2.1 Element ✓
- 2.1.2.2 Consists of only one type of atom. ✓
- 2.1.2.3 It is a combination of different metals (copper and zinc). ✓
- 2.2. Pots and pans are made of metal, because metal is a good conductor which allows heat to be transferred so that food can cook. ✓ The handles are insulators so that you do not burn your hands when you pick up a hot pot. ✓
- 2.3.1. (a) Temperature ✓
(b) Time ✓
- 2.3.2. Boiling point is the temperature of a liquid at which its vapour pressure equals to the external (atmospheric) pressure ✓✓
- 2.3.3. The temperature remains constant because the energy is absorbed by the molecules. ✓
The strength of the intermolecular forces between the molecules decreases. ✓
The particles move faster and spaces between the particles increases. ✓
The substance changes phase from liquid to gas ✓

[15]

Physical Science Chemistry Eng P2

**Wolfgang Glänzel, Henk F. Moed, Ulrich
Schmoch, Mike Thelwall**



Physical Science Chemistry Eng P2:

Technological Change and the Environment Arnulf Grübler, Nebojša Nakićenović, William D. Nordhaus, 2002 The 13 articles of this collection are the result of three workshops on induced technological change ITC held in 1997 1999 at the International Institute for Applied Systems Analysis in Laxenburg Austria where two of the editors teach the third teaches economics at Yale U The chapters consider topics that include the history of the ITC debate international perspectives various ITC modeling approaches evolutionary interpretation of innovation in medical history the relation of costs and performance of new technologies with carbon dioxide emission reduction and the place of ITC in the context of global climate change policy Annotation copyrighted by Book News Inc Portland OR Academic Science National Science Foundation (U.S.), 1982 **Advanced Engineering Chemistry** Manas Senapati, 2006-04 **Physical science and physical reality** Louis Osgood Kattsoff, 2013-11-11 *Accessions of Unlimited Distribution Reports*, 1975-01-03 **Scientific and Technical Aerospace Reports**, 1985 **The Chemical News and Journal of Physical Science**, 1903 **Nuclear Science Abstracts**, 1975 *Journals of the House of Representatives of New Zealand* New Zealand. Parliament. House of Representatives, 1915 **Report** New Zealand. Department of Education, 1914 *Springer Handbook of Science and Technology Indicators* Wolfgang Glänzel, Henk F. Moed, Ulrich Schmoch, Mike Thelwall, 2019-10-30 This handbook presents the state of the art of quantitative methods and models to understand and assess the science and technology system Focusing on various aspects of the development and application of indicators derived from data on scholarly publications patents and electronic communications the individual chapters written by leading experts discuss theoretical and methodological issues illustrate applications highlight their policy context and relevance and point to future research directions A substantial portion of the book is dedicated to detailed descriptions and analyses of data sources presenting both traditional and advanced approaches It addresses the main bibliographic metrics and indexes such as the journal impact factor and the h index as well as altmetric and webometric indicators and science mapping techniques on different levels of aggregation and in the context of their value for the assessment of research performance as well as their impact on research policy and society It also presents and critically discusses various national research evaluation systems Complementing the sections reflecting on the science system the technology section includes multiple chapters that explain different aspects of patent statistics patent classification and database search methods to retrieve patent related information In addition it examines the relevance of trademarks and standards as additional technological indicators The Springer Handbook of Science and Technology Indicators is an invaluable resource for practitioners scientists and policy makers wanting a systematic and thorough analysis of the potential and limitations of the various approaches to assess research and research performance *Principles of Chemical Engineering Processes* Nayef Ghasem, Redhouane Henda, 2025-03-31 Principles of Chemical Engineering Processes Material and Energy Balances continues to serve an essential text guiding students on the basic

principles and calculation techniques used in the field of chemical engineering and providing a solid understanding of the fundamentals of the application of material and energy balances This third edition has been updated to reflect advances in the field and feedback from professors and students Packed with illustrative examples and case studies this book Features learning objectives and homework problems in every chapter new material on software modeling and additional and enhanced solved examples and problems Discusses problems in material and energy balances related to chemical reactors and explains the concepts of dimensions units psychrometry steam properties and conservation of mass and energy Demonstrates how Python MATLAB and Simulink can be used to solve complicated problems of material and energy balances and now features an introduction to the basics of building Simulink models Demonstrates how Python and its libraries such as NumPy and SciPy can be used to solve complex problems in material and energy balances and introduces the basics of building models using Python frameworks similar to Simulink Shows how to solve steady state and transient mass and energy balance problems involving multiple unit processes and recycle bypass and purge streams Develops quantitative problem solving skills specifically the ability to think quantitatively including numbers and units the ability to translate words into diagrams and mathematical expressions the ability to use common sense to interpret vague and ambiguous language in problem statements and the ability to make judicious use of approximations and reasonable assumptions to simplify problems Offers educational software and sample tutorials and quizzes for download Aimed at both chemical engineering students and professionals this book helps readers understand how to calculate manage and apply the key ideas of material and energy use in chemical processes through real world examples Lecture slides and a solutions manual are available with qualifying course adoption

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