

Complete in the following table:

Correct shapes included  
although they're only necessary if  
specifically asked for.

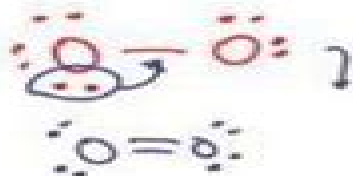
Group:	IA	IVA	VA	VIA	VIIA
Element:	H	C	N	O	F/Cl/Br/I
# valence electrons:	1	4	5	6	7
Normal # covalent bonds:	1	4	3	2	1

For the following neutral molecules, calculate the total number of valence electrons and draw the correct Lewis structure. (For neutral molecules, it's okay to assume the normal numbers of covalent bonds apply and that the central atom is the first non-H element in the formula.)

1.  $F_2$   $7 + 7 = 14$  ve total



2.  $O_2$   $6 + 6 = 12$  ve total



3.  $N_2$   $5 + 5 = 10$  ve total



4.  $CH_4$   $4 + 4(1) = 8$  ve total



5.  $CH_3Cl$   $4 + 3(1) + 7 = 14$  ve total



6.  $CH_2Cl_2$   $4 + 2(1) + 2(7) = 20$  ve total



# South Pasadena Bonding And Molecular Structure Worksheet Answers

**F Rizvi**



## **South Pasadena Bonding And Molecular Structure Worksheet Answers:**

Chemical Bonds Harry B. Gray, 1994-12-05 This profusely illustrated book by a world renowned chemist and award winning chemistry teacher provides science students with an introduction to atomic and molecular structure and bonding. This is a reprint of a book first published by Benjamin Cummings 1973      Bonding and Molecular Structure Russell S. Drago, 1971      **Structure of Molecules and the Chemical Bond** Y. K. Syrkin, M. E. Dyatkina, 1950      The Determination of Molecular Structure P. J. Wheatley, 1953      **The Determination of Molecular Structure** Peter Jaffrey Wheatley, 1962  
*The Determination of Molecular Structure* P. J. Wheatley, 1965      Molecular Structure and Energetics; Volume 1: Chemical Bonding Models, 1986      **Structure of Molecules and the Chemical Bond** J. K. Syrkin, Y. K. Syrkin, M. E. Dyatkina, 1964      **The Theory of Molecular Structure and Bonding**, 1980      MOLECULAR STRUCTURE AND BONDING; THE QUALITATIVE MOLECULAR ORBITAL APPROACH. B. M. GIMARC, 1979      **The Nature of the Chemical Bond and the Structure of Molecules and Crystals** Linus Pauling, 1963

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