

4 • Classifying Chemicals**1 • pH and pOH***Fill in the missing information:*

[H ⁺]	pH	pOH
1×10^{-5}	5	9
1×10^{-3}	3	12
1×10^{-12}	12	2
* 2.5×10^{-3}	2.6	11.4
1×10^{-6}	6	8
1×10^{-13}	13	1

use your calculator pH = $-\log(2.5 \times 10^{-3})$* **4 • Classifying Chemicals**2 • [H⁺] and [OH⁻]***Fill in the missing information:*

[H ⁺]	[OH ⁻]
1×10^{-4}	1×10^{-10}
1×10^{-6}	1×10^{-8}
1×10^{-10}	1×10^{-4}
1×10^{-12}	1×10^{-2}
2×10^{-6}	5×10^{-9}
3×10^{-8}	3.3×10^{-7}
4.8×10^{-3}	2.1×10^{-12}
6.25×10^{-14}	1.6×10^{-1}

Remember: $[H^{+}] \times [OH^{-}] = 1 \times 10^{-14}$

South Pasadena Ph And Poh Answers

ML Morrison



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