

2nd 6 weeks Study Guide Answers

Parts of the Atom

Label the parts of the atom.



Particle	Electron	Proton	Neutron
Location	Outside Nucleus	Nucleus	Nucleus
Charge	Negative	Positive	Neutral

Year	Scientist	Contribution
400 BC	Democritus	Based on the idea of an atom
1803	Dalton	Atoms are tiny particles
1897	Thomson	Plum pudding model Electrons
1911	Rutherford	Atoms have a central nucleus
1913	Bohr	Electrons revolve in the atom, small nucleus that is positively charged with a negative cloud of electrons circling it.

Process	Description	Advantages	Disadvantages
Observation	See, hear, feel, etc.	Simple, easy to do	Subjective, can be biased
Questioning	Ask questions about what you observe	Helps to identify the problem	Can lead to wrong conclusions
Hypothesis	A statement that can be tested	Can be tested and proven true or false	Can be wrong
Experiment	Test the hypothesis	Can be repeated and verified	Can be expensive and time-consuming
Conclusion	Summarize the results of the experiment	Can be used to make predictions	Can be biased
Communication	Share the results of the experiment with others	Can be used to make predictions	Can be biased

List and explain the steps of the scientific method

1. Observe something
2. Ask a question
3. Form a hypothesis
4. Research your topic & develop your experiment
5. Perform the experiment & test hypothesis
6. Collect Data
7. Analyze data, Conclude & Report

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Shaun Gabbidon, Helen Taylor Greene



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