

Name _____

LAB: TYPES OF CHEMICAL REACTIONS

PHYSICAL SCIENCE

Introduction

In this experiment, you will learn to differentiate among five general types of chemical reactions. (*Synthesis, Decomposition, Replacement, Single & Double, Combustion*) You will carry out certain representative reactions yourself, while your teacher will demonstrate another. Many chemical reactions are accompanied by observable physical changes. The appearance of a solid (precipitate) when solutions of substances are mixed is one such change. Others include the generation of a gas and a change in the color of a solution.

Materials

Magnesium	Potassium iodide	Potassium chromate
Hydrochloric acid	Sodium Bicarbonate	(tongs, alcohol lamp, test tube)
Lead (II) Nitrate	Silver nitrate	Watch glass

Procedure

Wear safety goggles and aprons during the entire lab.

1. **Magnesium metal and hydrochloric acid.** Place a small piece of magnesium on the watch glass. Add 10 drops of hydrochloric acid, and record observations on your data table. Dispose of the solid in the waste container and clean the watch glass.
2. **Magnesium metal and oxygen.** The teacher will perform this reaction. Record the results on your data table.
3. **Lead (II) Nitrate and potassium iodide.** Place two drops of lead (II) nitrate on the watch glass. Add two drops of potassium iodide. Record the results on your data table. Dispose of the materials, and clean the watch glass.
4. **Sodium bicarbonate with heat.** The teacher will perform this reaction. Record the results.
5. **Silver nitrate and potassium chromate.** Place two drops of silver nitrate on the watch glass. Add two drops of potassium chromate. Record the results in the data table. Dispose of the materials and clean the watch glass.
6. **Methane and oxygen.** This reaction occurs when a Bunsen burner is on. Come to the front of the classroom and demonstrate only with the teacher present.
7. Clean your table and return all equipment and materials.

Study Island Answers To Chemical Reactions Physical Science

Robert R. Redfield



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