# **Equilibrium position**



Chemists often refer to the position of equilibrium. The position of equilibrium refers to the side of the reaction arrow that has the greatest concentration at equilibrium.





Equilibrium to the left = when the concentration of the reactants is greater than the concentration of the products.

Equilibrium to the right =
when the concentrations of
the products is greater than
the concentration of the
reactants.

# **Teacher Edition 1percent Yield Study Guide Chemistry**

**Susan McMurry** 

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