Atomic and Ionic Radii Mr Cole Chemistry



Trends in Atomic Radii

1	Describe and explain the trend in atomic radii as you go down group 2:
2	Describe and explain the trend in atomic radii as you go across period 3.
s	Describe and explain the trend in atomic radii as you go across period 4 from Scandium to Zinc.
114	dii Comparisons
•	pare the radii of the following, including a description and explanation in your answer.
	a. Na and Na
	b. Li and O
-	

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