Name:

Energy and Thermal Energy Review

Thermal Energy	1. the sum of the kinetic and potential energy of the molecules of a substance
Conduction	2. transfer of heat by particles colliding with each other
Convection	3. transfer of heat by the movement of a gas or a liquid
Radiation	4. transfer of energy by electromagnetic waves
Thermometer	5. an instrument used to measure temperature
Conductor	6. a substance that conducts heat
Insulator	7. a material through which heat does not easily flow
Heat	8. thermal energy transferred because of a difference in temperature
Fahrenheit	9. temperature scale with 180 degrees between the freezing and boiling of H2O
Celcius	10. temperature scale with 100 degrees between the freezing and boiling of H2O
Temperature	11. the average kinetic energy of the particles of a substance
Specific Heat	12. amount of heat to raise 1 kg of a substance 1°C

13. As matter is heated, its particles move faster.

14. Particles that make up matter are in a state of constant motion.

The energy of moving particles is called kinetic energy.

The energy stored inside a battery is chemical energy.

Temperature is a measure of the average kinetic energy of atoms and molecules.

18. When a liquid is heated it will (expand or contract)? expand

19. A pot on a stove gets hot by means of conduction.

20. Since water has a higher specific heat than sand, water will heat slower than sand.

21. What are the energy transformations in the follow: a nuclear reaction generates heat to produce steam. The steam is used to turn a generator which finally produces electricity.

Nuclear to Thermal to Kinetic to Energy

22. Photosynthesis transforms electromagnetic energy from the Sun into chemical energy.

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Marcel A. Müller

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