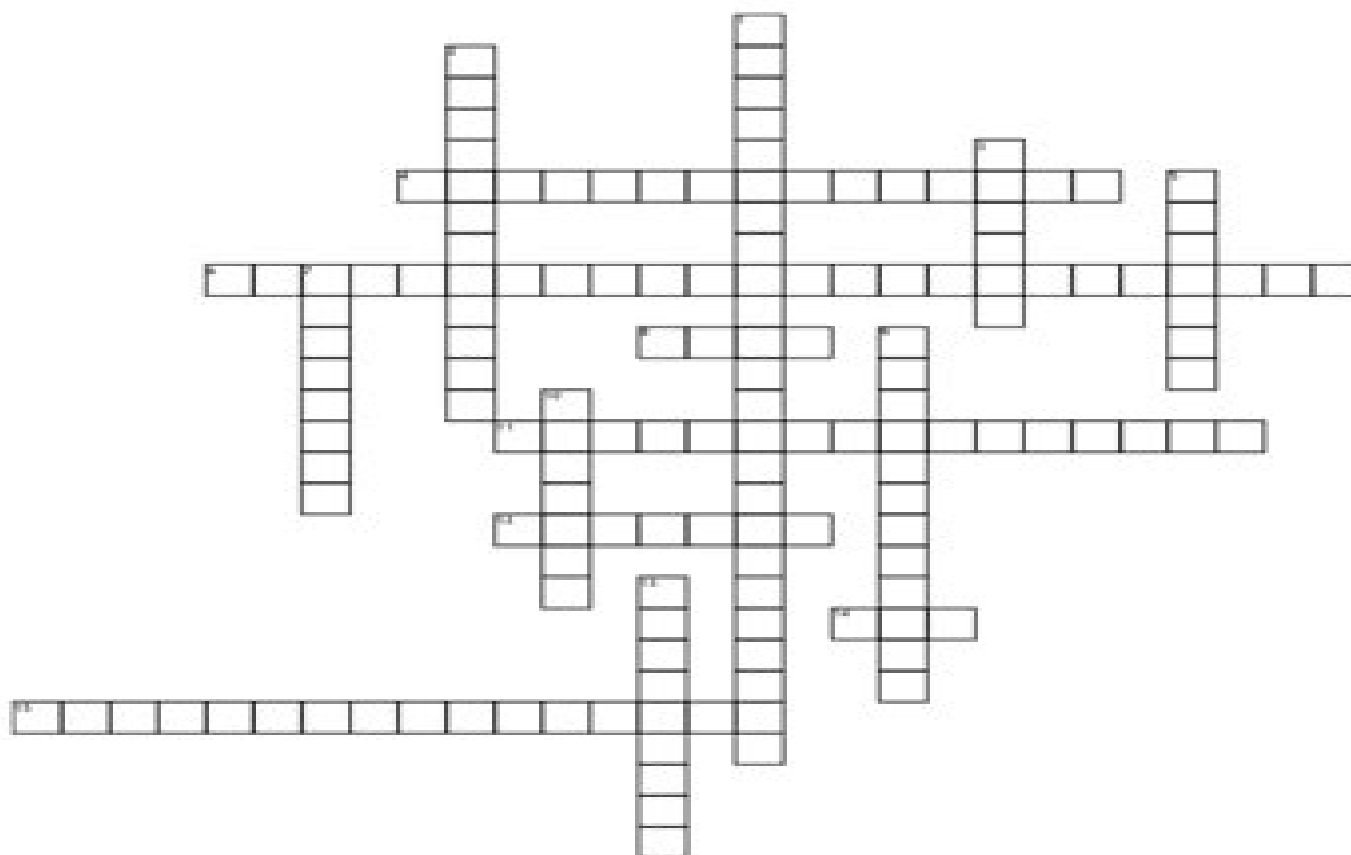


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Unit 8 Atoms & Chemical Bonds



Across

4. an attractive or repulsive force between two charged objects.
6. a property of matter that describe a substances based on its ability to change into a new substances with different properties while a physical property cannot be change
8. the basic unit of a chemical element
11. a definite quantity of electricity, either negative or positive, usually regarded as a more or less localized population of electrons separated or considered separately from their corresponding protons
12. an uncharged elementary particle that has a mass nearly equal to that of the proton and is present in all known atomic nuclei except the hydrogen nucleus.

14. an atom or group of atoms that carries a positive or negative electric charge as a result of having lost or gained one or more electrons.

15. unit of matter or energy that's the fundamental makeup of all matter. According to modern atomic theory, an atom has a nucleus, which is its center, or core.

Down

1. a property (as color, hardness, boiling point) of matter not involving in its manifestation a chemical change.
2. he chemical bond typical of the metallic state and characterized by mobile valence electrons that hold the atoms together usually in crystal lattices and are responsible for the good electrical and heat conductivity of metals.
3. a particle with a positive charge that is in the nucleus of an atom

5. any of two or more species of atoms of a chemical element with the same atomic number and nearly identical chemical behavior but with differing atomic mass or mass number and different physical properties.

7. an elementary particle consisting of a charge of negative electricity equal to about 1.602×10^{-19} coulomb and having a mass when at rest of about 9.109×10^{-31} kilogram or about $1/1836$ that of a proton.

9. bond formed by the sharing of a pair of electrons by two atoms.

10. composed of atoms that have the same atomic number, that is, each atom has the same number of protons in its nucleus as all other atoms

13. a chemical bond formed between oppositely charged species because of their mutual electrostatic attraction.

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