

Why aren't the elements listed in alphabetical order?

As charts go, the periodic table is a bit odd. It's not square. Large portions of the table appear to be missing at the top. It's not organized alphabetically so elements can be found easily. But to a chemist, the periodic table is a very powerful tool. The periodic table is organized by properties, both chemical and physical. Those properties relate to the electronic structure of the atoms of each element. In today's activity, you are going to study how the ground state electron configurations and structure of atoms are related to the shape and organization of the periodic table.

A blank periodic table grid with the following structure:

- s-block:** The first two columns (Groups 1 and 2).
- p-block:** The last six columns (Groups 13 to 18).
- d-block:** The ten columns between the s-block and p-block (Groups 3 to 10).
- f-block:** Two rows of seven columns each, located below the main grid, representing the lanthanide and actinide series.

The grid is labeled with periods 1 through 7 on the left side.

- ## Cracking the Periodic Table Code

# Teacher Guide Cracking The Periodic Table Code

**Clifford Lane**



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