

# FIRST GRADE PHONICS CURRICULUM

## LETTER REVIEW

**SMALL GROUP AND CENTER ACTIVITIES:** While students are working in their center groups, use the box to assign each student letter recognition and sound activities.

- Letter picture sort
- Spin & Color
- Day & Night Letter Cards
- Roll & Read Letter Recognition
- Letter Flip Book

**OBJECTIVES:**

- Recognize and name letters
- Identify the sounds of the letters

**MASTERS:**

- Letter recognition and sound
- Letter cards
- Letter sound cards
- Letter sound cards
- Letter sound cards
- Letter sound cards

### DAY 1

**PHONIC AWARENESS:** Sentence Segmentation - Flip out 5 envelopes or some other manipulatives to each student. Students will make an envelope for each word they hear in the following sentences.

I have two dogs.	Can I come with you?	Where are my glasses?
I can help you.	Can I have a drink?	How about the phone.

**LETTER REVIEW: a, A**

- Show the picture cards for each letter. Demonstrate how to write the uppercase and lowercase letters and how to make the sounds.
- Complete the sound sort together using the picture cards.
- Have students complete the handwriting and sound page for each letter.

### DAY 2

**PHONIC AWARENESS:** Sentence Segmentation - Flip out 5 envelopes or some other manipulatives to each student. Students will make a card for each word they hear in the sentences on the corresponding paper.

**LETTER REVIEW: a, p**

- Show the picture cards for each letter. Demonstrate how to write the uppercase and lowercase letters and how to make the sounds.
- Complete the sound sort together using the picture cards.
- Have students complete the handwriting and sound page for each letter.

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## LETTER REVIEW

### DAY 3

**PHONIC AWARENESS:** Sentence Segmentation - Flip out 5 envelopes or some other manipulatives to each student. Students will make an envelope for each word they hear in the following sentences.

My mom has two cats and one dog.	Will you help me clean my room?
I like to eat peanut butter and jelly.	Do you want some candy?

**LETTER REVIEW: c, C**

- Show the picture cards for each letter. Demonstrate how to write the uppercase and lowercase letters and how to make the sounds.
- Complete the sound sort together using the picture cards.
- Have students complete the handwriting and sound page for each letter.

### DAY 4

**PHONIC AWARENESS:** Sentence Segmentation - Flip out 5 envelopes or some other manipulatives to each student. Students will make a card for each word they hear in the sentences on the corresponding paper.

**LETTER REVIEW: s, S**

- Show the picture cards for each letter. Demonstrate how to write the uppercase and lowercase letters and how to make the sounds.
- Complete the sound sort together using the picture cards.
- Have students complete the handwriting and sound page for each letter.

### DAY 5

**PHONIC AWARENESS:** Sentence Segmentation - Flip out 5 envelopes or some other manipulatives to each student. Students will make a card for each word they hear in the following sentences.

I want to go home now.	I like your black dog run by.
Can you help me make my bed?	My sister is going to her friend's house.

**LETTER REVIEW: t, T**

- Show the picture cards for each letter. Demonstrate how to write the uppercase and lowercase letters and how to make the sounds.
- Complete the sound sort together using the picture cards.
- Have students complete the handwriting and sound page for each letter.

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# Teaching Phonics To First Grade Guide

**Christian G. Meyer**



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web objective to apply stoichiometric principles to calculate the appropriate chemical equation for the reaction between iron metal and a solution of copper i sulfate introduction in the form of finely split powder copper metal is produced as a byproduct of the process as iron ions the iron metal dissolves

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web purpose to enhance the understanding of stoichiometry a reaction between iron and copper ii sulfate solution will be conducted this will help you to differentiate limiting and excess reactant in a chemical reaction finally the theoretical and percent yield of this reaction will be calculated

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web purpose in this experiment you will observe a single replacement reaction you will also use stoichiometry to predict what the theoretical yield of product is and calculate a percent yield reaction iron copper ii sulfate copper iron ii sulfate pre lab questions write the balanced equation

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web october 3 2017 abstract the techniques used in this lab are quantitative transfer and vacuum filtration with the reaction of 8.001 grams of copper ii sulfate  $\text{CuSO}_4$  and 2.0153 grams of iron powder  $\text{Fe}$  the goal of this experiment was to determine the product of copper ii sulfate with iron

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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8.00 grams of copper ii sulfate onto a

piece of filter paper taking the filter paper's mass into account add these to the beaker 3 measure 50.0 ml of water in the cylinder and add it to the beaker 4

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web stoichiometry lab the reaction of iron with copper ii sulfate the study of stoichiometry deals with the calculation of quantities in a chemical reaction how much product will be produced how much reactant do you need to make that much product

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web description great lab for use during stoichiometry unit in chemistry reinforces ideas of mole ratios and stoichiometry calculations by measuring the mass of iron that reacts and the mass of copper metal produced students can calculate the ratio of moles of reactant to moles of product this experimental mole ratio from lab can be compared

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web in this experiment we will use stoichiometric principles to deduce the appropriate equation for the reaction between metallic iron and a solution of copper ii sulfate this reaction produces metallic copper which is seen precipitating as a finely divided red powder

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web in this lab we will use experimental evidence and the principles of stoichiometry to determine which ion is formed in a single replacement reaction between iron and copper ii sulfate pre lab assignment in preparation for the lab activity we will investigate another single replacement reaction between copper and silver nitrate

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web through this experiment we sought to use stoichiometric principles which derive from stoichiometry the study of substance quantities in ratio or proportion in a chemical reaction in this lab experiment the principles help distinguish the equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction

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web this experiment is focused on a reaction between iron metal and a solution of copper ii sulfate we are able to figure out the chemical equation between the two by using stoichiometric principles the iron displaces the copper ii sulfate ions the copper ii is then reduced and gains electrons

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web stoichiometry can be defined as the relationship between the relative quantities of substances taking part in a reaction or forming a compound stoichiometric principles can be used to deduce the appropriate chemical equation for a reaction in our experiment we used solid iron in the form of powder was added to a solution of copper ii sulfate

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web under the finalize of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away cooper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with a percent yield of 100 72 introduction

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web stoichiometry the reaction of iron with copper ii sulfate purpose to enhance the understanding of stoichiometry a reaction between iron and copper ii sulfate solution will be conducted this will help you to differentiate limiting and excess reactant in a chemical reaction

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web stoichiometry lab iron and copper sulfate sample data data and observations mass of copper ii sulfate pentahydrate 3 27 g of g mass of filter paper mass of filter paper and dry copper mass of dry copper 1 28 g

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web nov 23 2021 the electrical safety authority esa would like to advise that the government has approved the regulatory amendment to adopt the 2021 ontario electrical safety code oesc which comes into effect on may 5 2022 the new code will be available for order in december 2021 the hardcopy of the code book will be available

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web ontario regulation 777 21 electrical safety code the ontario electrical safety code ontario regulation 777 21 is comprised of the canadian electrical code part i cec together with specific ontario amendments development of the cec part i is led by

the canadian standards association csa

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web the 27th edition of the ontario electrical safety code oesc includes the 24th edition of the canadian electrical code part i plus ontario specific amendments the oesc is the law in ontario for electrical installation this new edition has been updated to enhance electrical safety and better protect electrical workers and the public

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web ontario amendments to the canadian electrical code part i c22 1 1 of 123 sections 0 through 86 of the canadian electrical code part i c22 1 21 exclusive of any references to any of the appendices is adopted as part of the electrical safety code with the following amendments section 0 object scope and definitions see appendix g

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web 1 2 18 oesc means the ontario electrical safety code incorporated by reference under ontario regulation 164 99 1 2 19  
ohsa means the occupational health and safety act r s o 1990 c o 1 1 2 20 primary distribution line

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government and consumer services has approved the regulatory amendment to adopt the 2021 ontario electrical safety code  
oesc which comes into effect on may 5 2022