

4) Oxidation-Reduction Reactions

Oxidation - is a loss of electrons
 - is an increase in oxidation number

Reduction - is a gain of electrons
 - is a decrease in oxidation number

Oxidizing Agent - brings about the oxidation of another species
↑ gets reduced

Reducing Agent - brings about the reduction of another species
↑ causes something else to gain electrons by losing electrons itself (Is Oxidized)
↑ gets oxidized

★ OIL RIG (Oxidation Is Loss; Reduction Is Gain)
 LEO-GER (Lose Electrons Oxidation ~ Gain Electrons Reduction)

Memorize common oxidizing pathways (reducing goes backward)

1. Metal $\xrightarrow{\text{ox}}$ metal ion
 $2n \rightarrow 2n^{2+}$
2. Metal $\xrightarrow{\text{ox}}$ metallic ion $\xrightarrow{\text{ox}}$ metallic ion
 $\text{Cu} \rightarrow \text{Cu}^+ \rightarrow \text{Cu}^{2+}$
3. Halide ion $\xrightarrow{\text{ox}}$ halogen $\xrightarrow{\text{ox}}$ halite ion $\xrightarrow{\text{ox}}$ halate ion (similar for $\text{NO}_2^- \xrightarrow{\text{ox}} \text{NO}_3^-$)
 $\text{Cl}^- \rightarrow \text{Cl}_2 \rightarrow \text{ClO}^- \rightarrow \text{ClO}_2^- \rightarrow \text{ClO}_3^- \rightarrow \text{ClO}_4^-$

Memorize common oxidizing / reducing agents (see bright orange sheet)

Ex. $\text{MnO}_4^- \rightarrow \text{Mn}^{2+}$ (acidified solution)
 $\text{MnO}_4^- \rightarrow \text{MnO}_2$ (basic solution)
 $\text{Cr}_2\text{O}_7^{2-} \rightarrow \text{Cr}^{3+}$ (acidified solution)

===Identify the substances being oxidized and reduced in each of the following.===



MEMORIZE!

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M Carnoy



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Dr Nandkishor Chandan, 2021-07-19 This book entitled OXIDATION AND REDUCTION REAGENTS IN ORGANIC SYNTHESIS has been particularly addressed to the graduate and postgraduate students who have opted for the Organic Chemistry study course as per the UGC syllabus This book is equally useful for those students who are preparing for the NET JRF CSIR SET SLET GATE NET ICAR and other competitive examinations like MPSC and UPSC This book includes two volumes which divided into four chapters as volume I cover oxidation and volume II covers reduction in which general methods of preparations synthetic applications and mechanism is discussed in details with different sets of examples The large numbers of problems with solutions have been included at the end of each reagent discussions and covering the questions asked in different universities and competitive examinations The organic synthesis is one of the most important branch of chemical science which wide exploited in the architecture of organic molecules with high biological significance

Oxidation-reduction Reactions Irish Science Teachers' Association. Chemistry Syllabus Committee, 1988* *The Electron in Oxidation-reduction* De Witt Talmage Keach, 1926

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Table of Contents Oxidation And Reduction Packet Answer Key

1. Understanding the eBook Oxidation And Reduction Packet Answer Key
 - The Rise of Digital Reading Oxidation And Reduction Packet Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Oxidation And Reduction Packet Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Oxidation And Reduction Packet Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Oxidation And Reduction Packet Answer Key
 - Personalized Recommendations
 - Oxidation And Reduction Packet Answer Key User Reviews and Ratings
 - Oxidation And Reduction Packet Answer Key and Bestseller Lists

5. Accessing Oxidation And Reduction Packet Answer Key Free and Paid eBooks
 - Oxidation And Reduction Packet Answer Key Public Domain eBooks
 - Oxidation And Reduction Packet Answer Key eBook Subscription Services
 - Oxidation And Reduction Packet Answer Key Budget-Friendly Options
6. Navigating Oxidation And Reduction Packet Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Oxidation And Reduction Packet Answer Key Compatibility with Devices
 - Oxidation And Reduction Packet Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Oxidation And Reduction Packet Answer Key
 - Highlighting and Note-Taking Oxidation And Reduction Packet Answer Key
 - Interactive Elements Oxidation And Reduction Packet Answer Key
8. Staying Engaged with Oxidation And Reduction Packet Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Oxidation And Reduction Packet Answer Key
9. Balancing eBooks and Physical Books Oxidation And Reduction Packet Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Oxidation And Reduction Packet Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Oxidation And Reduction Packet Answer Key
 - Setting Reading Goals Oxidation And Reduction Packet Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Oxidation And Reduction Packet Answer Key
 - Fact-Checking eBook Content of Oxidation And Reduction Packet Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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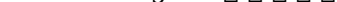
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