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CCCCO $\xrightarrow{\text{PCC}}$ CCCC=O

PCC - $\text{:}\ddot{\text{O}}\text{:} - \text{C}(=\text{O}) - \text{Cl}$

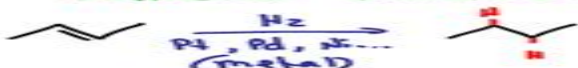


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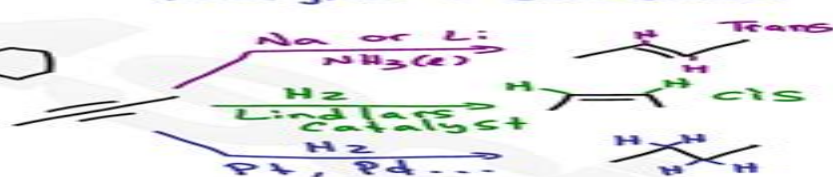
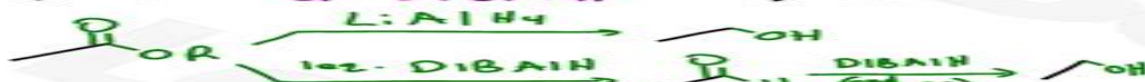
CC(=C)C $\xrightarrow[\text{1. KMnO}_4]{\text{O}_3, \text{CH}_2\text{I}_2}$ CC(=O)C=O $\xrightarrow[\text{2. H}_2\text{O}]{\text{hot}}$ CC(=O)C(=O)C

only 1° 2° + 3°
get oxidized
4° = N/A

Alkene Reduction

c1ccccc1>[Li] or [Na] / ROH, NH3>c1ccccc1

Alkyne Reduction

CC(=O)O>[LiAlH4]>CCO

Wolff - Kishner



Clemmenson



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