

ASSIGNING OXIDATION NUMBERS – WORKSHEET 1

Part A:

In the following questions, give the oxidation number of the indicated atoms/ion.

1. N in N_2O_3 _____

2. S in H_2SO_4 _____

3. C _____

4. C in CO _____

5. Na in NaCl _____

6. H in H_2O _____

7. Ba in BaCl_2 _____

8. N in NO_2^- _____

9. S in Al_2S_3 _____

10. S in HSO_4^- _____

11. Cl in $\text{Fe}(\text{ClO}_2)_3$ _____

12. Fe in $\text{Fe}(\text{ClO}_2)_3$ _____

13. N in NO_3^- _____

14. Cu^{2+} _____

15. Zn^{2+} _____

16. C in CH_4 _____

17. Mn in MnO_2 _____

18. S in SO_3^{2-} _____

19. Mg^{2+} _____

20. Cl^- _____

21. O_2 _____

22. P_4 _____

23. Na in Na_2S _____

24. S in H_2S _____

25. Ca^{2+} _____

26. C in CN^- _____

27. H in OH^- _____

28. Mn in KMnO_4 _____

29. I in $\text{Mg}(\text{IO}_3)_2$ _____

30. C in $\text{C}_2\text{O}_4^{2-}$ _____

Oxidation Numbers Answer Key

Greg Curran

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Oxidation Numbers Answer Key:

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