

# Answers: Oxidation Number

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1. N in  $\text{N}_2\text{O}_3$   $\text{N}^{3+}$
2. S in  $\text{H}_2\text{SO}_4$   $\text{S}^{6+}$
3. C  $\text{C}^0$
4. C in CO  $\text{C}^{2+}$
5. Na in NaCl  $\text{Na}^{+}$
6. H in  $\text{H}_2\text{O}$   $\text{H}^{+}$
7. Ba in  $\text{BaCl}_2$   $\text{Ba}^{2+}$
8. N in  $\text{NO}_2^-$   $\text{N}^{3+}$
9. S in  $\text{Al}_2\text{S}_3$   $\text{S}^{2-}$
10. S in  $\text{HSO}_4^-$   $\text{S}^{6+}$
11. Cl in  $\text{Fe}(\text{ClO}_2)_3$   $\text{Cl}^{3+}$
12. Fe in  $\text{Fe}(\text{ClO}_2)_3$   $\text{Fe}^{3+}$
13. N in  $\text{NO}_3^-$   $\text{N}^{5+}$
14.  $\text{Cu}^{2+}$   $\text{Cu}^{2+}$
15.  $\text{Zn}^{2+}$   $\text{Zn}^{2+}$
16. C in  $\text{CH}_4$   $\text{C}^{4+}$
17. Mn in  $\text{MnO}_2$   $\text{Mn}^{4+}$
18. S in  $\text{SO}_3^{2-}$   $\text{S}^{4+}$
19.  $\text{Mg}^{2+}$   $\text{Mg}^{2+}$
20.  $\text{Cl}^-$   $\text{Cl}^-$
21.  $\text{O}_2$   $\text{O}_2^0$
22.  $\text{P}_4$   $\text{P}_4^0$
23. Na in  $\text{Na}_2\text{S}$   $\text{Na}^{+}$
24. S in  $\text{H}_2\text{S}$   $\text{S}^{2-}$
25.  $\text{Ca}^{2+}$   $\text{Ca}^{2+}$
26. C in  $\text{CN}^-$   $\text{C}^{4+}$
27. H in  $\text{OH}^-$   $\text{H}^{+}$
28. Mn in  $\text{KMnO}_4$   $\text{Mn}^{7+}$
29. I in  $\text{Mg}(\text{IO}_3)_2$   $\text{I}^{5+}$
30. C in  $\text{C}_2\text{O}_4^{2-}$   $\text{C}^{2+}$

# Oxidation Numbers Section Review Answers Answers

**Barron's Educational Series, Albert S.  
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    - Chapter 3: Oxidation Numbers Section Review Answers Answers in Everyday Life
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    - Chapter 5: Conclusion
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  6. In chapter 5, the author will draw a conclusion about Oxidation Numbers Section Review Answers Answers. This chapter will summarize the key points that have been discussed throughout the book.
- This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Oxidation Numbers Section Review Answers Answers.

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