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Scott Brown



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web identify the missing particle in the following nuclear reaction $^1_0\text{n} + ^{235}_{92}\text{U} \rightarrow ^{140}_{54}\text{Xe} + ^{92}_{38}\text{Sr} + ^1_0\text{n}$

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web answer 4 correct in the reactants and products must be the same part b what is the value of in the following nuclear reaction $^{65}_{29}\text{Cu} + ^1_0\text{n} \rightarrow ^{65}_{28}\text{Ni} + ^1_0\text{n}$ express your answer as an integer hint 1 how to approach the problem use rule 2 the total of the subscripts atomic numbers mathematically this can be written as answer

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