

Use the partial periodic table to answer questions 4 -7.

6
C
Carbon
12.0107

— Atomic number

— Symbol

— Name

— Average Atomic Mass

**Partial Periodic Table of the Elements**

IA	1	H	Hydrogen	1.00794
2	Li	Lithium	6.941	
3	Na	Sodium	22.98976928	
4	K	Potassium	39.0983	

2	Be	Beryllium	9.012182
3	Mg	Magnesium	24.304
4	Ca	Calcium	40.078

IIA	13	B	Boron	10.811
IVA	14	C	Carbon	12.0107
V	15	N	Nitrogen	14.0064
VI	16	O	Oxygen	15.9994
VII	17	F	Fluorine	18.9984
VIIIA	18	Ne	Neon	20.1797
13	Al	Aluminum	26.9815385	
14	Si	Silicon	28.0855	
15	P	Phosphorus	30.973761508	
16	S	Sulfur	32.06	
17	Cl	Chlorine	35.453	
18	Ar	Argon	39.948	

### Question 5

Suppose scientists discovered four new elements (W, X, Y, Z) while studying rock and soil samples brought back from a Mars mission. Which Lewis dot structure represents an element that should be placed in column VIIA (17) of the periodic table?

- A.  B.  C.  D. 

# Partial Periodic Answer Physical Science Lab

**Wenbin Ji**



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