

# FTCE

## (PRE-KINDERGARTEN/PRIMARY PK-3 (053) EXAM) STUDY GUIDE & PRACTICE QUESTIONS 2025-2026

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web the name of a binary compound containing monatomic ions consists of the name of the cation the name of the metal followed by the name of the anion the name of the nonmetallic element with its ending replaced by the suffix ide some examples are given in table 2 6 names of some ionic compounds table 2 6 compounds containing

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molecular h 3 po 4 empirical h 3 po 4 [click here for a video of the solution](#)

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web bismuth has a molar mass of one mole of the compound would result in two moles of bismuth a total of 417.96g tellurium has a molar mass of one mole of the compound would result in three moles of tellurium a total of 382.8g add the mass of bismuth and the mass of tellurium per mole to find the total molar mass divide the mass of bismuth

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web in molecular compounds the atom binds each other through covalent bonds in salts it is held together with ionic bonds these are the two types of bonds out of which every compound is made of 2 example of compounds example of compounds includes water  $\text{H}_2\text{O}$  hydrogen peroxide  $\text{H}_2\text{O}_2$  etc

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web the substances mentioned above exemplify the two basic types of chemical compounds molecular and ionic methane and water are composed of molecules that is they are molecular compounds sodium chloride on the other hand contains ions it is [2.6 ionic and molecular compounds chemistry 2e openstax](#) - Oct 05 2022

web molecular compounds many compounds do not contain ions but instead consist solely of discrete neutral molecules these molecular compounds covalent compounds result when atoms share rather than transfer gain or lose electrons covalent bonding is an important and extensive concept in chemistry and it will be treated in considerable

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web may 28 2020 answer problem 4.3.1 3.4.3.1.3 for each of the following compounds state whether it is ionic or covalent if it is ionic write the symbols for the ions involved a.  $\text{NF}_3$  b.  $\text{BaO}$  c.  $\text{NH}_4\text{Cl}$  d.  $\text{SrH}_2$  e.  $\text{I}_2$  f.  $\text{Na}_2\text{O}$  answer a. answer b. answer c. answer d. answer e. answer f. problem 4.3.1 4.4.3.1.4

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web iron is an element it is made of only one type of atom iron atoms sulfur is another element it contains only sulfur atoms and nothing else so it is pure when we mix two different pure

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web jul 25 2022 write the name for each molecular compound co h 2s sf 6 n 2o 2 strategy first you have to know how to name a molecular compound to name a compound its the prefix name of first element prefix and name of the second element with the suffix ide the prefixes are as follows mono 1 di 2 tri 3 tetra 4

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web molecular compounds are inorganic compounds that take the form of discrete molecules examples include such familiar substances as water h<sub>2</sub>o h 2 o and carbon dioxide co<sub>2</sub> co 2 these compounds are very different from ionic compounds like sodium chloride nacl nacl

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web may 28 2020 what is the total mass of hydrogen in each of the molecules a ch 4 b chcl 3 c c 12 h 10 o 6 d ch 3 ch 2 ch 2 ch 2 ch 3 answer a answer b answer c answer d problem 4 4 1 3 calculate the molecular or formula mass of each of the following a p 4 b h 2 o c ca no 3 2 d ch 3 co 2 h acetic acid

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