

Name \_\_\_\_\_ Date \_\_\_\_\_ Class \_\_\_\_\_

## 17

## THERMOCHEMISTRY

**SECTION 17.1 THE FLOW OF ENERGY—HEAT AND WORK (pages 505–510)**

*This section explains the relationship between energy and heat, and distinguishes between heat capacity and specific heat.*

**► Energy Transformations (page 505)**

- What area of study in chemistry is concerned with the heat transfers that occur during chemical reactions? thermochemistry
- Where the use of energy is concerned (in a scientific sense), when is work done?  
Work is done when a force is used to move an object.
- Circle the letter next to each sentence that is true about energy:
  - ☒ Energy is the capacity for doing work or supplying heat.
  - ☒ Energy is detected only because of its effects.
  - ☐ Heat is energy that transfers from one object to another because they are at the same temperature.
  - ☒ Gasoline contains a significant amount of chemical potential energy.
- Circle the letter next to each sentence that is true about heat.
  - ☒ One effect of adding heat to a substance is an increase in the temperature of that substance.
  - ☐ Heat always flows from a cooler object to a warmer object.
  - ☒ If two objects remain in contact, heat will flow from the warmer object to the cooler object until the temperature of both objects is the same.

**► Exothermic and Endothermic Processes (pages 506–507)**

- What can be considered the “system” and what are the “surroundings” when studying a mixture of chemicals undergoing a reaction? Write your answers where indicated below.

System: The mixture of chemicals itself is considered the system.

Surroundings: Everything but the mixture of chemicals is the surroundings, but practically speaking, the immediate vicinity of the system.

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**L Towner**



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quality articles parse the latest supramolecular advancements and methods in the areas of chemistry biochemistry biology environmental and materials science and engineering physics computer science and applied mathematics Designed for specialists and students alike the set covers the fundamentals of supramolecular chemistry and sets the standard for relevant future research Environmental Science | AICTE Prescribed Textbook - English Subrat Roy,2021-11-01 Environmental

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Chemical Literacy and Writing Chemical Reactions Nikolay Gerasimchuk,2022-09-12 Writing chemical reactions in general and inorganic chemistry is not a trivial task However writing reactions for chemical processes correctly is a clear indicator of proficiency and competence in a subject Unfortunately very few students grasp the concept of the correct writing of chemical reactions quickly and so are unable to move through topics of general analytical and inorganic chemistry freely Because the ability to write and balance different types of chemical reactions is a fundamental issue this becomes a key

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*Hansen Solubility Parameters* Charles M. Hansen, 2007-06-15 Hansen solubility parameters HSPs are used to predict molecular affinities solubility and solubility related phenomena Revised and updated throughout Hansen Solubility Parameters A User s Handbook Second Edition features the three Hansen solubility parameters for over 1200 chemicals and correlations for over 400 materials including p **Journal of Education** ,1893 **British Books in Print** ,1978

Nanostructured Ceramic Oxides for Supercapacitor Applications Avinash Balakrishnan, K.R.V. Subramanian, 2014-02-19 A fresh and innovative technology is currently being recognized as a viable replacement for batteries Research in the field of supercapacitors as well as in the area of ceramic materials and their application to supercapacitor development has spawned

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