

Study Guide

Key Concepts



14.1 Properties of Gases

- Gases are easily compressed because of the space between particles in a gas.
- The amount of gas (n), volume (V), and temperature (T) are factors that affect gas pressure (P).



14.2 The Gas Laws

- As the pressure of a gas increases, the volume decreases, if the temperature is constant.
- As the temperature of an enclosed gas increases, the volume increases, if the pressure is constant.
- As the temperature of an enclosed gas increases, the pressure increases, if the volume is constant.

- The combined gas law allows you to do calculations for situations in which only the amount of gas is constant.



14.3 Ideal Gases

- To calculate the number of moles of a contained gas requires an expression that contains the variable n .
- Real gases differ most from an ideal gas at low temperatures and high pressures.



14.4 Gases: Mixtures and Movements

- In a mixture of gases, the total pressure is the sum of the partial pressures of the gases.
- Gases of lower molar mass diffuse and effuse faster than gases of higher molar mass.

Vocabulary

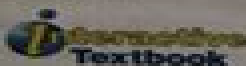
- Boyle's law (p. 418)
- Charles's law (p. 420)
- combined gas law (p. 424)
- compressibility (p. 413)
- Dalton's law of partial pressures (p. 432)
- diffusion (p. 435)
- effusion (p. 435)
- Gay-Lussac's law (p. 422)
- Graham's law of effusion (p. 435)
- ideal gas constant (p. 426)
- ideal gas law (p. 426)
- partial pressure (p. 432)

Key Equations

- Boyle's law: $P_1 \times V_1 = P_2 \times V_2$
- Charles's law: $\frac{V_1}{T_1} = \frac{V_2}{T_2}$
- Gay-Lussac's law: $\frac{P_1}{T_1} = \frac{P_2}{T_2}$
- Combined gas law: $\frac{P_1 \times V_1}{T_1} = \frac{P_2 \times V_2}{T_2}$
- Ideal gas law: $P \times V = n \times R \times T$ or $PV = nRT$
- Dalton's law: $P_{\text{total}} = P_1 + P_2 + P_3 + \dots$
- Graham's law: $\frac{\text{Rate}_A}{\text{Rate}_B} = \sqrt{\frac{\text{molar mass}_B}{\text{molar mass}_A}}$

Organizing Information

Use these terms to construct a concept map that organizes the major ideas of this chapter.



Interactive Textbook

Concept Map 14 Solve the Concept Map with the help of an interactive guided tutorial.

with ChemASAP

Boyle's law

Charles's law

partial pressure

combined gas law

diffusion

Prentice Hall Chemistry Chapter 14 Answers

Anil Kumar, Rakesh K. Gupta



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Chemistry of Aqueous Uranium (V) Solutions K. A. Kraus, F. Nelson, 1949 **Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science**, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction *Prentice Hall Chemistry*, 2000 Aqueous Chemistry of Ruthenium in the +2, +3, and +4 Oxidation States Howard Hamilton Cady, 1957 Forensic Chemistry Michael Grossman, 2021-12-20 FORENSIC CHEMISTRY FUNDAMENTALS strives to help scientists workplace health safety and other areas are included And two issues of Canadian legal process are argued as essays in the final two chapters **Study Guide and Solutions Manual to Accompany Fundamentals of Organic Chemistry** John McMurry, Susan McMurry, 1986 **Introduction to Quantum Mechanics in Chemistry** Mark A. Ratner, George C. Schatz, 2001 This book serves as a self study guide to familiarize users with the crucial language of modern chemistry science It provides a background of electronic structure programs and includes worked examples in problem solving and computer exercises For computational chemists materials scientists and chemical engineers who want to learn more about their field without unnecessary complexity detail or formalism **Catalog of Copyright Entries. Third Series** Library of Congress. Copyright Office, 1959 Includes Part 1 Number 2 Books and Pamphlets Including Serials and Contributions to Periodicals July December **Mathematical Methods for Physicists** George Brown Arfken, George B. Arfken, Hans J. Weber, Frank E. Harris, 2013 Table of Contents Mathematical Preliminaries Determinants and Matrices Vector Analysis Tensors and Differential Forms Vector Spaces Eigenvalue Problems Ordinary Differential Equations Partial Differential Equations Green's Functions Complex Variable Theory Further Topics in Analysis Gamma Function Bessel Functions Legendre Functions Angular Momentum Group Theory More Special Functions Fourier Series Integral Transforms Periodic Systems Integral Equations Mathieu Functions Calculus of Variations Probability and Statistics **The Handbook of Board Governance** Richard Leblanc, 2020-04-06 The revised new edition of the must read guide for executives provides comprehensive coverage of topics in corporate governance by leading subject matter experts The Handbook of Board Governance is the marketing leading text on public nonprofit and private board governance Providing comprehensive in depth coverage this unique text represents a collaboration of internationally recognized academics and prominent organization directors executives managers and advisors Contributors include Ariel Fromer Babcock Robert Eccles Alice Korngold Ellie Mulholland Michael Useem Elizabeth Valentine and John Zinkin Practical expert guidance enables readers to understand value creation and the strategic role of the board risk governance and oversight audit and compensation committee effectiveness CEO succession planning and other diverse board duties and

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Study Guide and Solutions Manual for Organic Chemistry Susan McMurry, John McMurry, 1988

Digital Simulation in Electrochemistry Dieter Britz, Jörg Strutwolf, 2016-05-09 This book explains how the partial differential equations pdes in electroanalytical chemistry can be solved numerically It guides the reader through the topic in a very didactic way by first introducing and discussing the basic equations along with some model systems as test cases systematically Then it outlines basic numerical approximations for derivatives and techniques for the numerical solution of ordinary differential equations Finally more complicated methods for approaching the pdes are derived The authors describe major implicit methods in detail and show how to handle homogeneous chemical reactions even including coupled and nonlinear cases On this basis more advanced techniques are briefly sketched and some of the commercially available programs are discussed In this way the reader is systematically guided and can learn the tools for approaching his own electrochemical simulation problems This new fourth edition has been carefully revised updated and extended compared to the previous edition

Lecture Notes in Physics Vol 666 It contains new material describing migration effects as well as arrays of ultramicroelectrodes It is thus the most comprehensive and didactic introduction to the topic of electrochemical simulation

Chemistry and Physics of Aqueous Gas Solutions William Alfred Adams, 1975

Modern Electrochemistry John O'M. Bockris, Amulya K.N. Reddy, 1973-10-31

7 The Electrified Interface

7 1 Electrification of an Interface

7 1 1 The Electrode Electrolyte Interface The Basis of Electrodics

7 1 2 New Forces at the Boundary of an Electrolyte

7 1 3 The Interphase Region Has New Properties and New Structures

7 1 4 An Electrode Is Like a Giant Central Ion

7 1 5 The Consequences of Compromise Arrangements The Electrolyte Side of the Boundary Acquires a Charge

7 1 6 Both Sides of the Interface Become Electrified The So Called Electrical Double Layer

7 1 7 Double Layers Are Characteristic of All Phase Boundaries

7 1 8 A Look into an El

Characterization of Cavitation Bubbles and Sonoluminescence Rachel Pflieger, Sergey I. Nikitenko, Carlos

Cairós, Robert Mettin, 2019-01-31 This book presents the latest research on fundamental aspects of acoustic bubbles and in particular on various complementary ways to characterize them It starts with the dynamics of a single bubble under ultrasound and then addresses few bubble systems and the formation and development of bubble structures before briefly reviewing work on isolated bubbles in standing acoustic waves bubble traps and multibubble systems where translation and interaction of bubbles play a major role Further it explores the interaction of bubbles with objects and highlights non spherical bubble dynamics and the respective collapse geometries It also discusses the important link between bubble dynamics and energy focusing in the bubble leading to sonochemistry and sonoluminescence The second chapter focuses on the emission of light by cavitation bubbles at collapse sonoluminescence and on the information that can be gained by sonoluminescence SL spectroscopy e g the conditions reached inside the bubbles or the nature of the excited species formed This chapter also includes a section on the use of SL intensity measurement under pulsed ultrasound as an indirect way to estimate bubble size and size distribution Lastly since one very important feature of cavitation systems is their sonochemical activity the final chapter presents chemical characterizations the care that should be taken in using them and the possible visualization of chemical activity It also explores the links between bubble dynamics SL spectroscopy and sonochemical activity This book provides a fundamental basis for other books in the Molecular Science Ultrasound and Sonochemistry series that are more focused on applied aspects of sonochemistry A basic knowledge of the characterization of cavitation bubbles is indispensable for the optimization of sonochemical processes and as such the book is useful for specialists

researchers engineers PhD students etc working in the wide area of ultrasonic processing **Chemistry** Theodore Lawrence Brown, 1988 Thermodynamics of Solutions Eli Ruckenstein, Ivan L. Shulgin, 2009-06-17 This book consists of a number of papers regarding the thermodynamics and structure of multicomponent systems that we have published during the last decade Even though they involve different topics and different systems they have something in common which can be considered as the signature of the present book First these papers are concerned with difficult or very nonideal systems i e systems with very strong interactions e g hydrogen bonding between components or systems with large differences in the partial molar volumes of the components e g the aqueous solutions of proteins or systems that are far from normal conditions e g critical or near critical mixtures Second the conventional thermodynamic methods are not sufficient for the accurate treatment of these mixtures Last but not least these systems are of interest for the pharmaceutical biomedical and related industries In order to meet the thermodynamic challenges involved in these complex mixtures we employed a variety of traditional methods but also new methods such as the fluctuation theory of Kirkwood and Buff and ab initio quantum mechanical techniques The Kirkwood Buff KB theory is a rigorous formalism which is free of any of the approximations usually used in the thermodynamic treatment of multicomponent systems This theory appears to be very fruitful when applied to the above mentioned difficult systems Fundamentals of Polymer Engineering, Revised and Expanded Anil Kumar, Rakesh K.

Gupta,2003-01-21 Exploring the characterization thermodynamics and structural mechanical thermal and transport behavior of polymers as melts solutions and solids this text covers essential concepts and breakthroughs in reactor design and polymer production and processing It contains modern theories end of chapter problems and real world examples for a clear understanding of polymer function and development Fundamentals of Polymer Engineering Second Edition provides a thorough grounding in the fundamentals of polymer science for more advanced study in the field of polymers Topics include reaction engineering of step growth polymerization emulsion polymerization and polymer diffusion *Ion-Selective*

Electrodes in Analytical Chemistry Henry Freiser,2012-12-06 Ion selective electrodes continue to be one of the more exciting developments in electro analytical chemistry in the last 10 years This is evidenced in the large and continually growing literature in the field It is important and necessary in such a rapidly growing area to be able to take stock i e to present a well rounded up to date review of important developments In this volume reviews by many of the leading practitioners and pioneers in this field contribute to what we consider to be a generous coverage of both fundamental aspects of ion selective electrodes and their applications to analytical chemistry Although this volume is not intended to be exhaustive we have attempted to produce a stand alone text dealing with all major current developments Indeed since some of the theoretical approaches are not yet universally agreed on each of the first five chapters deals with theory and principles of the nature and behavior of ion selective electrodes from the vantage point of the authors own experience and understanding In view of the rapid expansion of this field plans for future volumes are now being formulated Henry Freiser Tucson Arizona vii Contents Chapter 1 Theory and Principles of Membrane Electrodes R P Buck 1 Potential Generating Processes 1 1 1 Interfaces Fixed Charges Charged Sites and Charge Carriers 1 1 2 Ion Exchange as a Potential Generating Process 5 1 3 Diffusion and Migration 8 1 4 Electrochemical Potentials Fluxes and Mobility 10 1 5 *Physical Chemistry* Robert G.

Mortimer,2000-04-28 This new edition of Robert G Mortimer s *Physical Chemistry* has been thoroughly revised for use in a full year course in modern physical chemistry In this edition Mortimer has included recent developments in the theories of chemical reaction kinetics and molecular quantum mechanics as well as in the experimental study of extremely rapid chemical reactions While Mortimer has made substantial improvements in the selection and updating of topics he has retained the clarity of presentation the integration of description and theory and the level of rigor that made the first edition so successful Emphasizes clarity every aspect of the first edition has been examined and revised as needed to make the principles and applications of physical chemistry as clear as possible Proceeds from fundamental principles or postulates and shows how the consequences of these principles and postulates apply to the chemical and physical phenomena being studied Encourages the student not only to know the applications in physical chemistry but to understand where they come from Treats all topics relevant to undergraduate physical chemistry

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