

## Chapter 7 Chemical Reactions

6. What is a double-replacement reaction? A double-replacement reaction is a reaction in which two different compounds exchange positive ions and form two new compounds.
7. Complete the chart by filling in the general forms of the reactions shown.

General Forms	
Single-Replacement Reaction	Double-Replacement Reaction
$A + BC \longrightarrow A + BC$	$AB + CD \longrightarrow AD + CB$

8. Lead(II) nitrate reacts with potassium iodide to form lead(II) iodide and potassium nitrate. Write the balanced equation for this double-replacement reaction.  $Pb(NO_3)_2 + 2KI \longrightarrow PbI_2 + 2KNO_3$
9. Circle the letter of the correct answer. Calcium carbonate,  $CaCO_3$ , reacts with hydrochloric acid,  $HCl$ , in a double-replacement reaction. What are the products of this reaction?
- ☒ a. calcium chloride,  $CaCl_2$ , and carbonic acid,  $H_2CO_3$   
☐ b. calcium hydride,  $CaH_2$ , chlorine,  $Cl_2$ , and carbon dioxide,  $CO_2$   
☐ c. calcium hydrogen carbonate,  $Ca(HCO_3)_2$ , and chlorine,  $Cl_2$   
☐ d. calcium perchlorate,  $Ca(ClO_4)_2$ , and methane,  $CH_4$
10. Is the following sentence true or false? A combustion reaction is a reaction in which a substance reacts with carbon dioxide, often producing heat and light. false
11. Methane,  $CH_4$ , burns in oxygen to form carbon dioxide and water. Write the balanced equation for this reaction.  $CH_4 + 2O_2 \longrightarrow CO_2 + 2H_2O$
12. Is the following sentence true or false? The reaction that forms water can be classified as either a synthesis reaction or a combustion reaction. true

## Reactions as Electron Transfers (pages 204–205)

13. What is an oxidation-reduction reaction? An oxidation-reduction reaction is a reaction in which electrons are transferred from one reactant to another.
14. Calcium reacts with oxygen to form calcium oxide. Which reactant is oxidized in this reaction? calcium
15. Is the following sentence true or false? When calcium reacts with oxygen, each calcium atom gains two electrons and becomes a calcium ion with a charge of  $2+$ . false
16. Is the following sentence true or false? Oxygen must be present in order for an oxidation-reduction reaction to take place. false
17. The process in which an element gains electrons during a chemical reaction is called reduction.

# Prentice Hall Chemistry Ch 7 Answers

**Justin P. Lomont, Ian C. Stewart**



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**Computational Chemistry** Errol G. Lewars, 2010-11-10 This corrected second edition contains new material which includes solvent effects the treatment of singlet diradicals and the fundamentals of computational chemistry *Computational Chemistry Introduction to the Theory and Applications of Molecular and Quantum Mechanics* is an invaluable tool for teaching and researchers alike The book provides an overview of the field explains the basic underlying theory at a meaningful level that is not beyond beginners and it gives numerous comparisons of different methods with one another and with experiment The following concepts are illustrated and their possibilities and limitations are given potential energy surfaces simple and extended Hueckel methods ab initio AM1 and related semiempirical methods density functional theory DFT Topics are placed in a historical context adding interest to them and removing much of their apparently arbitrary aspect The large number of references to all significant topics mentioned should make this book useful not only to undergraduates but also to graduate students and

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*An Introduction to Spectroscopy, Atomic Structure and Chemical Bonding* Terry L. Meek, 1998 An Introduction to Spectroscopy presents the most fundamental concepts of inorganic chemistry at a level appropriate for first year students and in a manner comprehensible to them This is true even of difficult topics such as the wave mechanical atom symmetry elements and symmetry operations and the ligand group orbital approach to bonding The book contains many useful diagrams illustrating among other things the angular dependence of atomic wave functions the derivation of energy level diagrams for polyatomic molecules close packed lattices and ionic crystal structures The diagrams of the periodic variation of atomic and molecular properties showing trends across periods and down groups simultaneously are especially instructive Spectroscopy is presented mainly as a tool for the elucidation of atomic and molecular structures Each chapter begins with a clear and concise statement of What Every First year Student Should Know About outlining the background knowledge that the student is assumed to have from previous courses and thus pointing out what topics might need to be reviewed There are also detailed statements of the objectives of each chapter a number of worked examples interspersed in the text and a comprehensive set of problems and exercises to test the student s understanding Tables of data throughout the text and appendices at the end provide much valuable information

*Catalytic Oxidations with Hydrogen Peroxide as Oxidant* G. Strukul, 2013-03-14 Hydrogen peroxide is a chemical that is becoming increasingly fashionable as an oxidant both in industry and in academia and whose production is expected to increase significantly in the next few years This growth in interest is largely due to environmental considerations related to the clean nature of hydrogen peroxide as an oxidant its by product being only water To date this chemical has largely been employed as a non selective oxidant in operations like the bleaching of paper cellulose and textiles or in the formulation of detergents and only to a minimal extent in the manufacture of organic chemicals This book has been organized to cover the different aspects of the chemistry of hydrogen peroxide The various chapters into which the book is divided have been written critically by the authors with the general aim of stimulating new ideas and emphasizing those aspects that are likely to lead to new developments in organic synthesis in the coming future

**Structure and Physics of Viruses** Mauricio G. Mateu, 2024-12-31 The second edition of this book provides a completely updated account of the structure dynamics and physics of viral particles from the moment they emerge by self assembly from viral components produced in the infected cell through their extracellular stage until they recognize and infect a new host cell and cease to exist as they lose their physical integrity to initiate a new infectious cycle New insights

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*Surface Chemistry of Froth Flotation* S. Ramachandra Rao, 2013-06-29 th The technology of froth flotation invented in the early 20 century was first used for the concentration of sulfide minerals Since then it has been applied for the processing of many nonsulfide ores as well including oxides carbonates silicates soluble minerals like halite and sylvite and energy minerals like coal and bitumen In recent years it has been used for several nonmineral applications such as waste water treatment deinking of paper for recycling and resource recovery from industrial wastes the technology continues to grow with new applications reported every year Flotation is based on chemical phenomena occurring at the interfaces solid water and air water Surface Chemistry principles have played a significant role in the development of flotation technology Knowledge of aqueous solution chemistry and electrochemistry has added to our understanding of the reactions in flotation systems Professor Jan Leja s book has well served researchers and students as they tried to understand the chemistry of flotation and it is a significant contribution to the advancement of knowledge However since the book was first published new research techniques and ever growing information have made an update necessary The revised edition compiled by Dr S R Rao has brought together fundamental aspects of the chemistry of flotation and how they apply to practical systems It should serve all who are working in the area of flotation and interested in exploring new applications of flotation technology

*Metal Electrodeposition* Magdalena Nuñez, 2005 Electrochemistry is the branch of chemistry that deals with the chemical action of electricity and the production of electricity by chemical reactions In a world short of energy sources yet long on energy use electrochemistry is a critical component of the mix necessary to keep the world economies growing Electrochemistry is involved with such important applications as batteries fuel cells corrosion studies hydrogen energy

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