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Why is water a polar molecule? Water has H_2O , is a polar molecule.



Water is a polar molecule because it has a bent shape and a polar bond.

Bases

One way to familiar with water knowledge bases such as acids, bases, and salts. According to the Arrhenius theory, **bases** are ions compounds that dissociate into hydroxide ions (OH^-) when they dissolve in water. They are another example of strong electrolytes. For example, sodium hydroxide is an Arrhenius base that dissociates completely in water to give hydroxide ions (OH^-) and sodium ions (Na^+).

Most Arrhenius bases are formed from Group 1A (Li) and 2A (Ca) metals, such as $NaOH$, KOH , $LiOH$, and $Ca(OH)_2$. The hydroxide ions (OH^-) give Arrhenius bases common characteristics such as a bitter taste and a slippery feel. A base turns litmus indicator blue.

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