

Flame Test Lab



Objective

To prove that when heated elements emit energy that can be seen as light of different colors.

Materials

- Wooden splints
- Bunsen burner
- Striker
- Gas Source
- Googles
- Dipping Water
- Waste Water
- Weigh Boats
- Chemicals
 - Calcium Chloride
 - Strontium Chloride
 - Sodium Chloride
 - Copper Chloride
 - Potassium Chloride
 - Lithium Chloride

Procedure

1. Put on safety goggles.
2. Obtain: Bunsen Burner, Striker, 6 wooden splints, and 6 weigh boats with a very small amount of each of the sample chemicals.
3. Set up your bunsen burner and use the striker to carefully ignite the flame.
4. Dip one end of one of the wooden splints into the "dipping water" then into the first powdered chemical. The water ensures that the chemical sticks to the wood.

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Jessica J Manson

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Chemistry Dorin,Prentice Hall Dictionary Editors,1992 **Chemistry** Eugene LeMay, Jr.,Herbert Beall,Karen M. Robblee,Douglas C. Brower,Prentice-Hall Staff,2002-02 **Prentice Hall Chemistry** Antony C. Wilbraham,2008

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