

BALANCING NUCLEAR REACTIONS WORKSHEET

Predict the missing product or reactant in the following nuclear reactions. Determine the type of nuclear reaction (α emission, β emission, γ emission, positron emission, artificial transmutation, fission, or fusion) described. (30 points)

| Write the missing product or reactant | Type of Nuclear Reaction |
|---|--------------------------|
| 1. ${}_{19}^{42}\text{K} \rightarrow {}_{-1}^0\text{e} + \underline{\hspace{2cm}}$ | <hr/> |
| 2. ${}_{94}^{239}\text{Pu} \rightarrow {}_2^4\text{He} + \underline{\hspace{2cm}}$ | <hr/> |
| 3. ${}_{92}^{235}\text{U} \rightarrow \underline{\hspace{2cm}} + {}_{90}^{231}\text{Th}$ | <hr/> |
| 4. ${}_1^1\text{H} + {}_1^3\text{H} \rightarrow \underline{\hspace{2cm}}$ | <hr/> |
| 5. ${}_3^6\text{Li} \rightarrow {}_0^1\text{n} + {}_2^4\text{He} + \underline{\hspace{2cm}}$ | <hr/> |
| 6. ${}_{13}^{27}\text{Al} + {}_2^4\text{He} \rightarrow {}_{15}^{30}\text{P} + \underline{\hspace{2cm}}$ | <hr/> |
| 7. ${}_4^9\text{Be} + {}_1^1\text{H} \rightarrow \underline{\hspace{2cm}} + {}_2^4\text{He}$ | <hr/> |
| 8. ${}_{19}^{37}\text{K} \rightarrow {}_{+1}^0\text{e} + \underline{\hspace{2cm}}$ | <hr/> |
| 9. $\underline{\hspace{2cm}} + {}_0^1\text{n} \rightarrow {}_{56}^{142}\text{Ba} + {}_{36}^{91}\text{Kr} + 3{}_0^1\text{n}$ | <hr/> |
| 10. ${}_{92}^{238}\text{U} + {}_2^4\text{He} \rightarrow \underline{\hspace{2cm}} + {}_0^1\text{n}$ | <hr/> |
| 11. ${}_6^{14}\text{C} \rightarrow {}_7^{14}\text{N} + \underline{\hspace{2cm}}$ | <hr/> |
| 12. ${}_{75}^{187}\text{Re} + \underline{\hspace{2cm}} \rightarrow {}_{75}^{188}\text{Re} + {}_1^1\text{H}$ | <hr/> |
| 13. ${}_{11}^{22}\text{Na} + \underline{\hspace{2cm}} \rightarrow {}_{10}^{22}\text{Ne}$ | <hr/> |
| 14. ${}_{84}^{218}\text{Po} \rightarrow \underline{\hspace{2cm}} + {}_2^4\text{He}$ | <hr/> |
| 15. ${}_{99}^{253}\text{Es} + {}_2^4\text{He} \rightarrow {}_0^1\text{n} + \underline{\hspace{2cm}}$ | <hr/> |

Radioactivity And Nuclear Reactions Worksheet Key

Disha Experts



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Focus on physical science Charles H. Heimler, 1989

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Table of Contents Radioactivity And Nuclear Reactions Worksheet Key

1. Understanding the eBook Radioactivity And Nuclear Reactions Worksheet Key
 - The Rise of Digital Reading Radioactivity And Nuclear Reactions Worksheet Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Radioactivity And Nuclear Reactions Worksheet Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Radioactivity And Nuclear Reactions Worksheet Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Radioactivity And Nuclear Reactions Worksheet Key

- Personalized Recommendations
 - Radioactivity And Nuclear Reactions Worksheet Key User Reviews and Ratings
 - Radioactivity And Nuclear Reactions Worksheet Key and Bestseller Lists
5. Accessing Radioactivity And Nuclear Reactions Worksheet Key Free and Paid eBooks
 - Radioactivity And Nuclear Reactions Worksheet Key Public Domain eBooks
 - Radioactivity And Nuclear Reactions Worksheet Key eBook Subscription Services
 - Radioactivity And Nuclear Reactions Worksheet Key Budget-Friendly Options
 6. Navigating Radioactivity And Nuclear Reactions Worksheet Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Radioactivity And Nuclear Reactions Worksheet Key Compatibility with Devices
 - Radioactivity And Nuclear Reactions Worksheet Key Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Radioactivity And Nuclear Reactions Worksheet Key
 - Highlighting and Note-Taking Radioactivity And Nuclear Reactions Worksheet Key
 - Interactive Elements Radioactivity And Nuclear Reactions Worksheet Key
 8. Staying Engaged with Radioactivity And Nuclear Reactions Worksheet Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Radioactivity And Nuclear Reactions Worksheet Key
 9. Balancing eBooks and Physical Books Radioactivity And Nuclear Reactions Worksheet Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Radioactivity And Nuclear Reactions Worksheet Key
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Radioactivity And Nuclear Reactions Worksheet Key
 - Setting Reading Goals Radioactivity And Nuclear Reactions Worksheet Key
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Radioactivity And Nuclear Reactions Worksheet Key

- Fact-Checking eBook Content of Radioactivity And Nuclear Reactions Worksheet Key
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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