

## Radioactive Decay Worksheet #2

### Isotopes

- For atoms of the two different isotopes Lithium-6 and Lithium-7...
  - ...they must have the same number of what type of subatomic particle?
  - ...they must have different numbers of what type of subatomic particle?
- For each of the following isotopes, write in any missing information about that particular isotope.

Name	Oxygen-15	$^{58}_{28}\text{Ni}$	Zirconium- ____	Argon- ____	____ - ____	____ - ____
# of protons					9	30
# of neutrons			50	22	10	34

- Potassium (K) has three naturally-occurring isotopes, Potassium-39, Potassium-40, and Potassium-41. Using a periodic table, which isotope do you think is the most abundant in nature?

What leads you to this conclusion?

### Decay Reactions

Fill in the blank in each of the following decay reactions with the correct decay particle or decayed nucleus that will balance the decay reaction, and also state whether it is alpha or beta decay.



- Write out the decay reaction for a nucleus of Lead-210 decaying by alpha decay.

# Radioactive Decay Worksheet 2 Answers

**J Elliott**



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## **Table of Contents Radioactive Decay Worksheet 2 Answers**

1. Understanding the eBook Radioactive Decay Worksheet 2 Answers
  - The Rise of Digital Reading Radioactive Decay Worksheet 2 Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Radioactive Decay Worksheet 2 Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Radioactive Decay Worksheet 2 Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Radioactive Decay Worksheet 2 Answers
  - Personalized Recommendations
  - Radioactive Decay Worksheet 2 Answers User Reviews and Ratings
  - Radioactive Decay Worksheet 2 Answers and Bestseller Lists
5. Accessing Radioactive Decay Worksheet 2 Answers Free and Paid eBooks
  - Radioactive Decay Worksheet 2 Answers Public Domain eBooks
  - Radioactive Decay Worksheet 2 Answers eBook Subscription Services
  - Radioactive Decay Worksheet 2 Answers Budget-Friendly Options

6. Navigating Radioactive Decay Worksheet 2 Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Radioactive Decay Worksheet 2 Answers Compatibility with Devices
  - Radioactive Decay Worksheet 2 Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Radioactive Decay Worksheet 2 Answers
  - Highlighting and Note-Taking Radioactive Decay Worksheet 2 Answers
  - Interactive Elements Radioactive Decay Worksheet 2 Answers
8. Staying Engaged with Radioactive Decay Worksheet 2 Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Radioactive Decay Worksheet 2 Answers
9. Balancing eBooks and Physical Books Radioactive Decay Worksheet 2 Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Radioactive Decay Worksheet 2 Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Radioactive Decay Worksheet 2 Answers
  - Setting Reading Goals Radioactive Decay Worksheet 2 Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Radioactive Decay Worksheet 2 Answers
  - Fact-Checking eBook Content of Radioactive Decay Worksheet 2 Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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