LAB 9:

RADIOACTIVITY AND RADIATION

NOTE TO STUDENTS: This is a dry lab. You are only expected to perform the portions of this lab that do not require you to use laboratory equipment or supplies. When appropriate, sample data may be supplied in the lab's data tables to help you answer the questions.

Goals

- Understand that a very large number of random events becomes predictable
- Perform a simulation decay experiment
- Observe how a radioactive daughter product behaves

Materials and Equipment

50 small cubes (dice) 50 small spheres Foam cup, 8 oz Goggles

Materials Not Included

50 dimes (or pennies) Graph paper

Introduction

A radioactive material is an element that is unstable and emits a particle or ray of energy from its nucleus in an attempt to become stable. Radiation is what the element emits.

A half-life is the amount of time over which the radioactive element changes half of itself. Some unstable elements emit the radiation quickly, meaning they have short half-lives, and some emit over a long time, or have long half-lives. Whether for a short or long half-life, the process of emitting radiation is a very random. However, a very large number of random events can become predictable. That is where half-life enters. Sometimes the product of radioactive decay, called a daughter product, is unstable and radioactive. In fact, there can be an entire series of

radioactive decays before a stable product is found. This experiment is a simulation of a radioactive event where the daughter product is also radioactive. The dimes represent the radioactive parent, and the cubes represent the radioactive daughter. Since the above two objects have a different number of sides, they simulate different half-lives. The spheres represent a stable end product.

Procedure

- Look at the cubes and observe that they have dots on six sides. The side with two dots will indicate the cube has experienced radioactive decay.
- Count out 50 dimes and place them in the cup.
- Holding one hand over the cup, shake the cup and pour the dimes onto the table.
- Count and remove the dimes that landed with the heads side facing up. Replace those with a cube. Record the number of dimes remaining and the number of cubes in the Questions section.
- Repeat Procedures 3 and 4, but also remove all cubes that land with the two dots side up and replace them with a sphere.

Note: Make sure you have a total of 50 items, whether dimes, cubes, or spheres, to represent atoms.

- Do the above procedures for a total of 20 trials.
- 7. Record all results in Table 9.1.

Radioactivity Physics Lab Report

VM Jensen

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