#### + WS 6.2 - Nuclear Reactions

Complete each of the following nuclear reactions by determining the missing particle, then name that particle ("alpha particle" or "uranium-233", etc...) #1 is an example...

1. 14C> 14N + 9e beta particle	symbol	name
6 7 [-/	4He+≥ or α	alpha particle (a helium nucleus)
2 + .9e> +0Ar	oe orβ	beta particle
3. 202C1> + 195Mo + 41n	Oe +1	positron (a positive electron?t)
	ån	neutron
4. PBe +> 12C + in	1p or 1H+1	proton (hydrogen-1 nucleus)
5. gLi + 12C> 12F +		deutron (hydrogen-2 nucleus)

	-				
7. gHe	+ 3He	 ₹Be-	+		

15. Write the reaction for a Rn-224 atom undergoing 1 beta decay: What is the daughter?

16. Write the reaction for a Po-218 atom undergoing 1 alpha decay: What is the daughter?

17. What particle is produced when a U-234 atom undergoes a series of two alpha decays and three beta decays?

18. Pb-210 undergoes a series of alpha and beta decays to end up eventually as Ir-198. How many alpha particles and how many beta particles were emitted in all? Hint: do alpha's first, then beta's.  $\alpha = \beta = \beta$ 

# **Radioactivity And Nuclear Reactions Worksheet Answers**

Holt Rinehart & Winston, Holt, Rinehart and Winston Staff

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