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AN Whitehead



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Tracy Martin,2005 *Ford Vans Automotive Repair Manual* Ralph Rendina,John Harold Haynes,1995 Ford Full Size Vans
1992 95 Shop ManualHaynes32 pgs 76 ill *Industrial Education Magazine* Charles Alpheus Bennett,William Thomas
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web a write a balanced equation for the reaction $\text{Al} + \text{O}_2 \rightarrow \text{Al}_2\text{O}_3$ b determine the theoretical yield of Al_2O_3 $\frac{160.0 \text{ g Al}}{27 \text{ g/mol}} \times \frac{2 \text{ mol Al}_2\text{O}_3}{4 \text{ mol Al}} \times 101.96 \text{ g/mol} = 239.8 \text{ g}$ c determine the percent yield $\frac{260.0 \text{ g}}{239.8 \text{ g}} \times 100 = 108.4\%$

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